

Method Development and Applications for *In Situ* Rb-Sr Isotopic Analyses by Femtosecond LA-MC-ICPMS/MS

HAYWARD MELTON¹, TENG EE (TONY) YAP¹, ALICIA MARIE CRUZ-URIBE² AND FRANÇOIS L.H. TISSOT¹

¹California Institute of Technology

²University of Maine

The Rb-Sr beta-decay system is a fundamental tool in geochronology and geoforensics, widely applied to terrestrial, lunar, and Martian samples [1–3]. This technique is based on the radioactive decay of ⁸⁷Rb to ⁸⁷Sr and takes advantage of the substitution of Rb and Sr for K and Ca in rock-forming minerals which often have a range of Rb/Sr ratios. Over time, the decay of ⁸⁷Rb to ⁸⁷Sr increases the ⁸⁷Sr/⁸⁶Sr ratio, allowing us to construct an isochron and determine the crystallization age of a sample and its initial ⁸⁷Sr/⁸⁶Sr ratio.

Traditional Rb-Sr dating requires sample processing to mitigate the ⁸⁷Rb-⁸⁷Sr isobaric interference before analysis [4]. We develop a method integrating a femtosecond (fs) ESL NWRfemto 257nm LA system with a Thermo Fisher Scientific™ Neoma MC-ICPMS/MS, featuring a dual Wien mass filter and a collision-reaction cell for online Rb-Sr separation [5-7]. Using SF₆ gas, we selectively fluorinate Sr, shifting it by +19 m/z, and optimize key laser parameters.

We analyze reference glasses from NIST, MPI-DING, and USGS, covering a broad range of Rb-Sr compositions. Our results show ⁸⁷Sr/⁸⁶Sr accuracies exceeding 0.03% and ⁸⁷Rb/⁸⁶Sr accuracies within 5%, with precisions of ~1‰ and ~3% (2SE), respectively. Even with the benefits of an fs-LA system [8] this is still an unexpected degree of elemental fractionation which is likely attributable to transport effects.

Our method advances high-precision *in situ* Rb-Sr dating with broad applicability across geologic and planetary sciences. We demonstrate its potential for dating terrestrial feldspars and micas, tracing turquoise trade routes, and analyzing Martian meteorites, a subset of which will be presented.

References: [1] Papanastassiou D.A. & Wasserburg G.J. (1970) *EPSL*, 8, 269-278. [2] Jäger E. & Hunziker J.C. (1979) *Lectures in Isotope Geology*, 13-26. [3] Borg L.E. et al. (1999) *Science*, 286, 90-94. [4] Vanhaecke F. et al. (2001) *Fresenius J. Anal. Chem.*, 371, 915-920. [5] Bevan D. et al. (2021) *JAAS*, 36, 917-931. [6] Dauphas N. et al. (2022) *JAAS*, 37, 2420-2441. [7] Cruz-Urbe A.M. et al. (2023) *Geostand. Geoanalytical Res.*, 47, 795-809. [8] Ingo, H. and F. Von Blanckenburg (2007) *Spectro. Acta Part B*, 62, 410-422.