

# Ferric iron systematics of coexisting silicate and oxide minerals in hydrated metaperidotites with progressive subduction

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Hydrated metaperidotites convey large amounts of mineral-lattice bound water to subduction zone depths exceeding ~ 50 km [1] and exhibit a high capacity to oxidize other reservoirs, quantified by their redox budget [2]. With progressive subduction, hydrated metaperidotites undergo dehydration, and current models postulate a dominant partitioning of the redox budget into the dehydration fluid which is thought to represent the oxidizing agent for arc magmas [3]. However, recent studies challenge this paradigm, stating that the oxidation state of subducted hydrated metaperidotites is commonly too low to stabilize oxidized species in the dehydration fluid [4, 5].

To address this controversy surrounding redox budget cycling, we have undertaken a comprehensive study on  $\text{Fe}^{3+}$  –  $\text{Fe}^{2+}$  systematics in bulk rocks and silicate mineral separates formed upon metaperidotite dehydration, because iron is the most abundant, redox-sensitive element in these rocks. Metaperidotite samples from the central Swiss Alps were characterised in detail for their petrogenesis and preserve mineral assemblages formed upon antigorite breakdown (olivine + chlorite + orthopyroxene ± magnetite ± rare hematite). Antigorite dehydration product rocks are further distinguished into three groups based on  $f\text{O}_2$  (expressed relative to the QFM buffer): highly (QFM + 3.5), moderately (QFM + 3.0) and weakly oxidized (QFM + 1.0). Bulk rock  $^{57}\text{Fe}$  Mössbauer spectroscopy data reveal that from highly to weakly oxidized conditions, 35 – 4% of the total Fe is  $\text{Fe}^{3+}$ , with silicate minerals hosting 6 – 100% of the bulk  $\text{Fe}^{3+}$ . Mössbauer data on mineral separates identify chlorite as the main silicate  $\text{Fe}^{3+}$  carrier with 35%  $\text{Fe}^{3+}$  in one highly oxidized sample and 15%  $\text{Fe}^{3+}$  in one moderately oxidized sample, before orthopyroxene (~ 2%  $\text{Fe}^{3+}$ ) and olivine (< LOD – 1%  $\text{Fe}^{3+}$ ). Following antigorite dehydration, a large fraction of the redox budget hence remains rock-bound and is subducted further, possibly accounting for comparatively oxidised source domains in the convecting mantle.

## References:

- [1] Ulmer & Trommsdorff (1995) *Science*, 268(5212), 858–861
- [2] Evans (2006) *Geology*, 34(6), 489
- [3] Birner et al. (2017) *Journal of Petrology*, 58(9), 1755–