

# Rapid and ultrasensitive electrochemical detection of ammonia nitrogen in groundwater by a core-shell $\text{Cu}_2\text{O}@ \text{UiO}-66$ modified electrode

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Ammonia nitrogen ( $\text{NH}_3$  and  $\text{NH}_4^+$ ,  $\text{NH}_4\text{-N}$ ), as an important chemical indicator reflecting the quality of the water environment, plays a vital role in ecosystem stability. Especially in complex and variable low-quality groundwater environments, it is necessary to develop high-sensitive and selective  $\text{NH}_4\text{-N}$  detections to provide technical support for investigating its migration and remediation mechanism. The most-used detection method for  $\text{NH}_4\text{-N}$  is Nessler's reagent method, which suffers from a highly toxic substance, interference with other components, as well as only being used on-site instead of in-situ. Thus, the urgent key issue in this field is to develop a highly selective, environmentally friendly, and rapid in-situ detection method for ammonia nitrogen in groundwater where multiple components coexist.

In this study, a three-electrode detection system for  $\text{NH}_4\text{-N}$  with core-shell MOF ( $\text{Cu}_2\text{O}@ \text{UiO}-66$ ) as the working electrode was constructed. The functionalized  $\text{Cu}_2\text{O}@ \text{UiO}-66$  was synthesized by a facile hydrothermal method, followed by an in-situ growth on conductive carbon cloth (CC) to fabricate an electrochemical sensor ( $\text{Cu}_2\text{O}@ \text{UiO}-66/\text{CC}$ ). Under optimized conditions, the  $\text{Cu}_2\text{O}@ \text{UiO}-66/\text{CC}$  electrochemical sensor demonstrated a response time of only 12 seconds for detecting ammonia nitrogen through cyclic voltammetry (CV), with a low detection limit of  $0.231 \mu\text{M}$ , sensitivity of  $5.75 \mu\text{A} \cdot \mu\text{M}^{-1}$ , strong anti-interference ability, and a long service life. The detection mechanism was investigated as the presence of oxygen vacancies created by carboxylate ions and the nanoconfinement effect resulting from the pores of UiO-66 in the “shell” structure, which captures more ammonia nitrogen to the “core”  $\text{Cu}_2\text{O}$  to complete the subsequent electrocatalytic oxidation. Spiked water and real groundwater samples analysis showed that the sensor exhibited reliable recoveries (98.47% to 100.39%) and relative standard deviation (RSD) ( $<3\%$ ). This study offers a novel approach for swift and accurate in-situ detection of ammonia nitrogen in groundwater, indicating a great potential for practical application.

## References:

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