

The origin of sulfur in Canary Island magmas and its implications for Earth's deep sulfur cycle

ZOLTÁN TARACSÁK¹, MARGARET E. HARTLEY², RAY
BURGESS², MARIE EDMONDS¹, MARC-ANTOINE
LONGPRÉ³, BRIAN DAVID MONTELEONE⁴, ROMAIN
TARTÈSE² AND ALEXANDRA (SASHA) V. TURCHYN¹

¹University of Cambridge

²University of Manchester

³City University of New York

⁴Woods Hole Oceanographic Institution

Oxygen fugacity (fO_2) controls whether sulfur is present as sulfide or sulfate in silicic magmas: increasing fO_2 can lead to an order of magnitude increase in sulfur solubility [1]. This is reflected in the sulfur content of comparatively reduced and oxidised melts erupting at mid-ocean ridges (0.1-0.15 wt%) and at subduction-related volcanic arcs (up to 0.7 wt%), respectively [2]. Some ocean islands basalts (OIBs), including those erupted at El Hierro in the Canary Islands, reach S contents similar to subduction-related magmas [3]. While at subduction zones elevated sulfur contents and fO_2 is explained by slab-mantle interaction [1], the cause of sulfur enrichment and oxidation at ocean islands is more enigmatic.

We use sulfur isotope ($\delta^{34}S$) microanalyses of melt inclusions and glasses alongside bulk $\delta^{34}S$ analyses of lavas from El Hierro to investigate the origin of observed sulfur enrichment. Melt inclusions have $\delta^{34}S$ values between -1.6‰ and +4.1‰ (V-CDT), and S contents between 200 and 5300 $\mu\text{g/g}$, while lavas have $\delta^{34}S$ between -15.5 and +1.4‰ and S contents <400 $\mu\text{g/g}$. Using degassing and melting models we estimate undegassed melt $\delta^{34}S$ (between $0 \pm 2\text{‰}$) and mantle source S content ($310 \pm 120 \mu\text{g/g}$) for El Hierro magmas. On average, 60% of the sulfur in the mantle source is of recycled origin. To satisfy radiogenic isotope constraints on the fraction of recycled material in the mantle source (<10% [5]) we estimate that the recycled component should contain $\geq 1800 \mu\text{g/g}$ sulfur. A silica-rich partial melt of recycled oceanic crust could reach these modelled sulfur contents if it was partially oxidised ($\text{Fe}^{3+}/\Sigma\text{Fe}$ of >0.15). Based on our results we propose that recycling of Fe^{3+} in silicates is the source of elevated fO_2 in OIBs, while silicate melts that contain S^{6+} offer an effective pathway for oxidation in the upper mantle via melt-rock reactions.

References:

- [1] Jugo, Wilke & Botcharnikov (2010) *Geochimica et Cosmochimica Acta* 74(20), 5926-5938
- [2] Muth & Wallace (2022) *Earth and Planetary Science Letters* 599, 117836.
- [3] Longpré, Stix, Klügel & Shimizu (2017) *Earth and Planetary Science Letters* 460, 268-280.
- [4] Day, Pearson, Macpherson, Lowry & Carracedo (2010) *Geochimica et Cosmochimica Acta* 74(22), 6565-6589.