Rapid methane clumped isotope analysis using mid-infrared laser spectroscopy: development and first applications

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Methane clumped isotopes ($\Delta^{13}\text{CH}_3\text{D}$ and $\Delta^{12}\text{CH}_2\text{D}_2$), defined by deviations in the abundances of two doubly substituted isotopologues of CH₄, i.e., $^{13}\text{CH}_3\text{D}$ and $\Delta^{12}\text{CH}_2\text{D}_2$, from their stochastic distributions, serve as proxies for CH₄ formation temperatures or the contributions of kinetically controlled processes[1], among other applications. However, their low natural abundance poses analytical challenges: current methods either require \geq 20 hours per measurement with HR-IRMS[2] or \geq 20 mL STP of sample gas using laser spectroscopy[3].

To address these limitations, we developed a spectroscopic platform using quantum cascade laser absorption spectroscopy (QCLAS), with optimized spectral windows for clumped isotope analysis and a custom-built gas inlet system[4]. This technique allows for reducing sample size down to 3–7 mL CH₄ gas while achieving precision levels comparable to HR-IRMS. Samples larger than 10 mL can be quantified in a single run within 20 minutes.

With this novel approach, we analyze CH₄ from a variety of laboratory experiments and natural settings to explore possible applications. In a comprehensive technical study, we optimized the selection and use of cryogenic adsorbents for CH₄ storage for preservation of both bulk and clumped isotopic signatures. As a geological application, we analyzed thermogenic CH₄ trapped in source rock pore space to assess whether the CH₄ has recorded its formation history. Additionally, clumped isotope signatures of bubble gases collected from mud volcanoes and hybrid systems in Italy and Indonesia provide valuable insights into understanding microbial contributions to the initial thermogenic hydrocarbon sources. We also investigated the clumped isotopic fingerprint of CH₄ formed via iron-oxide-mediated reactions involving methyl radicals, a potential abiotic source of C1 and C2 compounds in nature[5]. Our presentation will showcase the potential of the spectroscopic technique as a highly practical tool for methane clumped isotope analysis.

[1] Young et al. (2017), Geochim. Cosmochim. Acta. 203, 235-264.; [2] Eldridge et al. (2019), ACS Earth Space Chem. 3(12), 2747-2764; [3] Gonzalez et al. (2019), Anal. Chem. 91(23),