

Photocatalytic H₂ production in natural fluid inclusions during Raman spectroscopic analyses

HAOKUN LIAN AND XIAOLIN WANG

Nanjing University

Hydrogen (H₂) is hailed as the key energy source of the 21st century, while natural H₂, characterized by its abundant reserves, low cost, cleanliness, and renewability, is referred to as “golden hydrogen”¹. However, the migration and accumulation mechanisms of natural H₂ are still poorly understood, and the exploration and development of natural H₂ reservoirs remain in their infancy². Mineral inclusions serve as natural time capsules, preserving evidence of fluid activities. Raman spectroscopy is the most reliable method to determine H₂ in single inclusion at present. H₂'s high reactivity and diffusivity preclude its long-term preservation in minerals, yet some reported H₂-bearing inclusions exhibit formation ages conflicting with theoretical preservation periods, which may lead to misinterpretations of H₂-bearing fluid activities. In this study, we conducted petrography and Raman spectroscopy on natural fluorite- and dolomite-hosted inclusions from the Ediacaran Dengying Formation in the Sichuan Basin, South China, containing CH₄+pyrobitumen±CO₂±H₂S±C₂H₆. We demonstrate the formation of H₂ within natural CH₄- and pyrobitumen-bearing fluid inclusions during Raman spectroscopic analyses, and exclude the potential for H₂ generation from pyrobitumen (Fig. 1). Raman spectroscopy temperature calibration and direct measurement by focusing the laser on a thermocouple show that the reaction temperature is approximately 32 °C. Our results indicate that, under 532 nm laser irradiation, pyrobitumen catalyzes CH₄ cracking, producing H radicals that rapidly recombine into H₂ within seconds to tens of seconds. If CO₂ exists in the inclusions, the H₂ yields is higher (11.5–43.0 mol%), while CO₂ is consumed and CO is produced (Fig. 1). This highlights the necessity of excluding artificially generated H₂ during spectral collection when studying the generation, migration, and accumulation of natural H₂ via fluid inclusion observations, to ensure the primary origin of H₂. Furthermore, considering that current H₂ production industries often face issues such as high reaction temperatures, catalyst deactivation, and significant greenhouse gas emissions, this low-temperature (32 °C) and rapid photocatalytic H₂ production may offer a novel perspective for industrial H₂ generation.

1. E. Hand, Hidden hydrogen. *Science*. **379**, 630–636 (2023).
2. Zgonnik, The occurrence and geoscience of natural hydrogen: A comprehensive review. *Earth-Sci. Rev.* **203**, (2020).

