Fractionation of K isotopes in coastal groundwater systems: Implications for the global K cycle and oceanic isotope budgets

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The coastal groundwater system plays a pivotal role in the global hydrological and biogeochemical cycles, with groundwater-derived solutes (e.g., nitrogen, phosphorus) estimated to be equivalent to that of the river input into the ocean^[1-2], thereby influencing the chemical composition of oceans. To date, little is known about potassium (K) fate in coastal aquifers and the impact of submarine groundwater discharge (SGD) on seawater δ^{41} K values, which are ~0.55% enriched in δ^{41} K relative to the bulk silicate Earth. Here, we present the K isotope and hydrochemical data of recharging river water, as well as unconfined and confined groundwater in the coastal region of Jiangsu province, China. The K isotope signatures revealed distinct fractionation mechanisms, with δ^{41} K values exhibiting an overall increase in unconfined aquifers (-0.41% to 0.78%) and confined aquifers (-0.14% to 1.37%). Specifically, chemical weathering and clay adsorption processes collectively govern the K concentrations and $\delta^{41}K$ values of nonseawater-influenced groundwater. In contrast, in seawaterinfluenced groundwater, the formation of clay minerals that sequester K from aguifers results in an elevation of δ^{41} K values to 0.78%-1.37%, as increased ionic strength inhibit K adsorption onto clays. Thus, a clay incorporation-induced K isotopic fractionation factor (α=0.9997 to 0.9999) was derived using the Rayleigh fractionation model. Utilizing a K massbalance model in conjunction with Monte Carlo simulations, we evaluated the response of seawater $\delta^{41}K$ values to SGD. Our results indicate that the δ^{41} K value of seawater increases by more than 0.3‰ when the K flux from SGD is comparable to the riverine input. This study highlights the role of SGD in the global K cycle and the oceanic isotopic budget, with important implications for the temporal evolution of seawater K isotope values.

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