Sub-nanometer resolution of zirconium isotopes in zircon and baddeleyite by Atom Probe Tomography

MICHELLE L FOLEY¹, MAURICIO IBANEZ-MEJIA¹, ELIAS BLOCH¹, STEPHAN GERSTL² AND FRANÇOIS L.H. TISSOT³

¹University of Arizona ²ETH Zürich ³Caltech Presenting Author: michellefoley@arizona.edu

The mechanisms behind zirconium isotope fractionation in igneous systems at the atomic scale are actively debated. At magmatic temperatures, the equilibrium fractionation of Zr between zircon and melt determined by ab initio calculations and zircon growth experimental studies is small ($\Delta^{94/90}$ Zr zircon-melt ~0.05 ‰ [1, 2]). However, several studies have documented significant mass-dependent variability in natural systems, which would require much larger fractionation factors with $\Delta^{94/90}$ Zr zircon-melt values up to +1.0 % [3]. Recent theoretical calculations [4] as well as zircon-growth experimental studies have suggested that these large fractionations must be driven by kinetic processes occurring at the liquid-solid interface [2]. To evaluate these nanoscale processes, we employ Atom Probe Tomography (APT) that utilizes point-projection Time-of-Flight (TOF) mass spectrometry to provide isotopic compositions of mineral samples with volumes on the order of $<0.02 \ \mu m^3$. In this study, we analyzed 12 samples by APT, including zircon samples from the Mud Tank carbonatite (MTUR1 crystal; [5]), as well as baddeleyite samples from the Duluth Gabbro (FC-1) and the Kovdor carbonatite. The bulk Zr isotope composition of the Mud Tank zircon has been well established [2,5], and as part of this study we provide new reference 94/90Zr values for baddeleyite crystals studied by APT. The foremost challenge for determining Zr isotope compositions from APT is assigning a background correction model to the TOF spectra. To evaluate the best method for reproducibly determining isotopic ratios via APT, we employ multiple background correction models and use different Zr±O species identified from the TOF spectra. The ability to quantify isotope ratios using APT is significant for our ability to identify mechanisms driving isotopic variations and mineral-melt isotope partitioning at the atomic scale.

[1] Méheut et al., 2021. Geochimica et Cosmochimica Acta 292, 217-234.

[2] Tompkins et al., 2023. Geochemical Perspective Letters 25.

[3] Ibañez-Mejia and Tissot, 2019. Science Advances 5, 1-14.

[4] Bindeman and Melnik, 2022. *Geochemical Perspective Letters* 24.

[5] Tompkins et al., 2020. J. Anal. At. Spectrom. 35.