## Characterizing the clumped isotope composition of microbially-produced H<sub>2</sub>

## NOAM LOTEM<sup>1</sup>, JARED LEADBETTER<sup>1</sup>, ELLIOTT P. MUELLER<sup>1</sup>, NAMI KITCHEN<sup>1</sup>, ITAY HALEVY<sup>2</sup> AND JOHN EILER<sup>1</sup>

<sup>1</sup>California Institute of Technology

<sup>2</sup>Weizmann Institute of Science

Presenting Author: nlotem@caltech.edu

Discerning microbially produced H<sub>2</sub> from abiotic H<sub>2</sub> could enhance our ability to identify H2 sources in the search of naturally occurring H<sub>2</sub> as an energy resource, and to study microbial modifications of H<sub>2</sub> stored in engineered crustal reservoirs. The molecular average deuterium content of H<sub>2</sub> has been used previously for identifying its sources and sinks, and recent studies suggest the multiply-substituted ('clumped') isotopologue, D2, could provide additional constraints on molecular formation mechanisms and temperatures [1, 2]. Here we present the first clumped isotope measurements of biologically produced H<sub>2</sub>. We measured the molecular average and clumped isotope composition of H<sub>2</sub> formed during fermentation in batch cultures of Clostridium pasteurianum; the H<sub>2</sub> was purified cryogenically and measured by the Thermo Fisher 253 Ultra high resolution IRMS. Our results show that microbially-produced H<sub>2</sub> expresses a consistent deviation from the equilibrium fractionation during exponential growth, for both the clumped and molecular average isotopic compositions, whereas during the stationary phase it conforms to nearequilibrium isotopic compositions at the culturing temperature. The specific combination of clumped and molecular average D contents we observed during the growth phase differs from equilibrium at any temperature and from simple kinetic fractionations such as diffusion; we suggest it may be a biologically regulated kinetic isotope signature unique to microbial growth. The shift from kinetic to equilibrium controls of isotopic composition moving from growth to stationary phase is consistent with theoretical calculation of the reversibility of the H<sub>2</sub> production reaction, because H<sub>2</sub> concentrations increase over growth, reaching and maintaining high levels during the stationary phase. It is plausible that in the environment, under lower H<sub>2</sub> concentrations, we would observe the distinctive, kinetic endmember during biological H<sub>2</sub> production. Since previous studies demonstrate that the clumped isotope composition of H<sub>2</sub> from natural reservoirs and hydrothermal vents reflects equilibrium values [2], the deviation from equilibrium during microbial growth can potentially be used to detect biological sources of subsurface H<sub>2</sub>.

[1] Popa, M.E. et al. (2019) Rapid Communications in Mass Spectrometry 33(3), 239-251.

[2] Mangenot, X. et al. (2023), Chemical Geology 621, 121278.