

## Computational Chemistry Applied to Solutions: Anions. II.

CORY C PYE

Saint Mary's University

Presenting Author: cory.pye@smu.ca

Computational chemistry has been employed by the author to interpret or predict the aqueous vibrational spectra of cations (Li<sup>+</sup>[1], Cd<sup>2+</sup>[2], Mg<sup>2+</sup>[3], Zn<sup>2+</sup>[4], Sc<sup>3+</sup>[5], Al<sup>3+</sup>[6], Ga<sup>3+</sup>[7], In<sup>3+</sup>[8], Bi<sup>3+</sup>[9], Be<sup>2+</sup>[10], Hg<sup>2+</sup>, Tl<sup>3+</sup>[11], Cu<sup>+</sup>[12], Pb<sup>2+</sup>[13], Sn<sup>2+</sup>[14], and Sb<sup>3+</sup>[15]), anions (SO<sub>4</sub><sup>2-</sup>[16], PO<sub>4</sub><sup>3-</sup>[17], HPO<sub>4</sub><sup>2-</sup>[18], H<sub>2</sub>PO<sub>4</sub><sup>-</sup>[19], ClO<sub>4</sub><sup>-</sup>, BrO<sub>4</sub><sup>-</sup>, SeO<sub>4</sub><sup>2-</sup>, AsO<sub>4</sub><sup>3-</sup>, VO<sub>4</sub><sup>3-</sup>[20], the borates[21], HSO<sub>4</sub><sup>-</sup>[22]; HSeO<sub>4</sub><sup>-</sup>[23]; HAsO<sub>4</sub><sup>2-</sup>[24]; HVO<sub>4</sub><sup>2-</sup>[25]); and complexes (LiX (X = F - I)[26], ScCl<sub>m</sub><sup>(3-m)+</sup>[27], ZnCl<sub>m</sub><sup>(2-m)+</sup>[28], ZnBr<sub>m</sub><sup>(2-m)+</sup>[29], CuCl<sub>m</sub><sup>(m-1)-</sup>[30]).

I will discuss microhydration of selected anions and illustrate the relationships between the spectra of anions in the gaseous, aqueous, and solid phases.

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