

Sulfur mass-independent fractionation in SO₂ photolysis: comparison between absorption spectra measurements and photolysis experiments of SO₂ - *GSJ Medal* *Lecture*

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Mass-independent fractionation of sulfur isotopes (MIF-S; non-zero $\Delta^{33}\text{S}$ values) in Archean sedimentary rocks will provide a hint of atmospheric composition at that time. However, the origin of MIF-S is still debated. SO₂ photolysis experiments produced large MIF-S signatures, and the results were highly sensitive to the experimental conditions [1]. Understanding the MIF-S mechanism in SO₂ photolysis is essential to explain the Archean MIF-S signatures quantitatively.

Isotope fractionation in SO₂ photolysis is predicted by absorption spectra of SO₂ isotopologues. Here, we measured the photoabsorption cross-sections of ³²SO₂, ³³SO₂, ³⁴SO₂, and ³⁶SO₂ from 206 to 220 nm at 296 K [2]. The spectral resolution of the present study ($\sim 1\text{ cm}^{-1}$) is 25 times higher than that of previous studies [3,4]. SO₂ exhibit complex ro-vibrational structures in their UV absorption spectrum; SO₂'s absorption causes MIF-S so-called "self-shielding" by slowing photolysis rate of only ³²SO₂, which is the most abundant isotopologue. Using our new spectra, we calculated the sulfur isotope fractionation for self-shielding during SO₂ photolysis. The calculated fractionation factors are compared with isotope fractionation factors observed in previous SO₂ photolysis experiments. We discuss the link of MIF-S between photolysis experiments and spectroscopic measurements. Additionally, we introduce some advances in understanding MIF-S mechanisms and modeling Archean atmospheric chemistry after the paper [2] was published. We performed MIF-S in SO₂ photolysis experiments in low-temperature/pressure atmospheres [5] and are developing a vertical atmospheric model with sulfur isotopes based on MIF-S observed in photolysis experiments. Finally, we summarize experimental and modeling studies of Archean MIF-S.

[1] Farquhar et al. (2001). *J. Geophys. Res.* 106(E12), 32829–32839.

[2] Endo et al. (2022). *Geochem. J.* 56(1), 40–56.

[3] Danielache et al. (2008). *J. Geophys. Res.* 113, D17314.

[4] Endo et al. (2015). *J. Geophys. Res. Atmos.* 120,