

Uranium isotope fractionation in the Paleoproterozoic ocean: A multi-core study of the Rove and Virginia Formations, Superior Province, North America

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The uranium isotope system is commonly used for reconstructing Phanerozoic and Proterozoic global ocean redox conditions. In non-euxinic settings, sedimentary $\delta^{238}\text{U}$ (Equation 1) is closer to coeval seawater ($\delta^{238}\text{U}_{\text{sw}}$) whereas reduction and burial of isotopically heavy U in euxinic settings produces black shales with significantly higher $\delta^{238}\text{U}$ than $\delta^{238}\text{U}_{\text{sw}}$. Areas where U reduction occurs are limited in the modern ocean, thus, modern $\delta^{238}\text{U}_{\text{sw}}$ (-0.4‰) is not significantly lower than continental input (-0.3‰). In widely anoxic oceans, global $\delta^{238}\text{U}_{\text{sw}}$ should be lower. Recently, Proterozoic application of this proxy has been questioned because other geochemical data suggest largely anoxic deep oceans, but $\delta^{238}\text{U}$ in some Proterozoic carbonates approach continental inputs, suggesting different and/or muted U isotope fractionation mechanism(s) compared to the present.

To better understand U isotope behavior in the Paleoproterozoic oceans, we present U isotope data from black shales of the ~ 1.83 Ga Rove/Virginia formations from four locations in the Animikie Basin. The shales have a wide range of $\delta^{238}\text{U}$ (-0.45‰ to $+0.43\text{‰}$). Generally, $\delta^{238}\text{U}$ correlates positively with local redox indicators (e.g., Fe speciation, U, V, Mo, Re, organic carbon). Euxinic samples record the highest $\delta^{238}\text{U}$ ($0.08 \pm 0.14\text{‰}$), whereas ferruginous ($-0.06 \pm 0.10\text{‰}$) and non-anoxic samples have lower $\delta^{238}\text{U}$ ($-0.24 \pm 0.15\text{‰}$) (Figure 1), consistent with modern U isotope systematics. We estimate that $\delta^{238}\text{U}_{\text{sw}}$ was between -0.8‰ and -0.4‰ using the above average $\delta^{238}\text{U}$ values and sediment–seawater offsets of $0.6\text{--}0.8\text{‰}$ and $<0.4\text{‰}$ in locally euxinic and ferruginous environments, respectively, as observed in modern settings. In one core interval, higher $\delta^{238}\text{U}$ ($0.28 \pm 0.14\text{‰}$) may have been produced by dynamic redox conditions, a phenomenon also observed in Recent organic-rich sediments. A seawater–sediment isotope offset in these environments closer to the theoretical maximum of 1.2‰ gives a $\delta^{238}\text{U}_{\text{sw}}$ estimate from the dynamic Rove interval that aligns with the low end of the aforementioned range of $\delta^{238}\text{U}_{\text{sw}}$. We conclude that at least some U isotope fractionation mechanisms observed today occurred in the Paleoproterozoic

oceans and that global ocean redox models using U isotopes should consider large fractionations in dynamic redox environments.

Equation 1

$$\delta^{238}\text{U} (\text{‰}) = \left(\frac{^{238}\text{U}/^{235}\text{U}_{\text{sample}}}{^{238}\text{U}/^{235}\text{U}_{\text{CRM145}}} - 1 \right) \times 1000$$

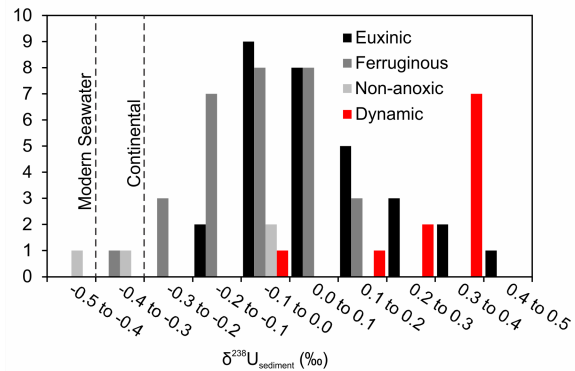


Figure 1. Distribution of uranium isotope compositions in sediments ($\delta^{238}\text{U}_{\text{sediment}}$) of the Rove and Virginia formations deposited under various redox conditions (defined by Fe speciation). A 'dynamic' interval spanning 30 cm in one core contains samples of non-anoxic, ferruginous, and euxinic deposition, and typically records higher $\delta^{238}\text{U}_{\text{sediment}}$ than other intervals.