

# Equilibrium europium isotope fractionation in igneous and metamorphic systems

EDWIN A. SCHAUBLE

UCLA

Presenting Author: [schauble@ucla.edu](mailto:schauble@ucla.edu)

Nuclear volume and mass-dependent isotope fractionations of europium isotopes are estimated using data from Mössbauer spectroscopy, NRIXS spectroscopy, and electronic structure calculations. The results indicate that  $\text{Eu}^{2+}$ -bearing species will have approximately 1‰ higher  $^{153}\text{Eu}/^{151}\text{Eu}$  than  $\text{Eu}^{3+}$ -bearing species in equilibrium at 25°C (298 K), and about 0.3‰ higher  $^{153}\text{Eu}/^{151}\text{Eu}$  at 700°C (973 K). The nuclear volume effect goes in the opposite direction from mass-dependent fractionation, and is predicted to dominate over mass dependent fractionation for most materials, especially at metamorphic and igneous temperatures.  $^{153}\text{Eu}/^{151}\text{Eu}$  fractionation appears to be primarily dependent on the oxidation state of europium. Estimated nuclear-volume dominated fractionations favoring high  $^{153}\text{Eu}/^{151}\text{Eu}$  in minerals that preferentially incorporate  $\text{Eu}^{2+}$ , such as plagioclase, are broadly consistent with recent measurements on igneous rocks showing low  $^{153}\text{Eu}/^{151}\text{Eu}$  in samples with strongly negative europium anomalies [1]. Plagioclase-melt fractionation is predicted to be most pronounced at intermediate oxygen fugacities, where  $\text{Eu}^{3+}$  is the main oxidation state in the melt or bulk rock but plagioclase mostly incorporates  $\text{Eu}^{2+}$ . The results are also consistent with the recent finding that equilibrium fractionation cannot explain cosmochemical REE fractionations observed in primitive meteoritical materials [2], because inclusion of the nuclear volume effect still yields net fractionations that are too small (~0.2‰ or less) at temperatures >1200 K where vapor-phase REE species are relevant. Similar techniques applied to estimate nuclear volume fractionation for  $^{142}\text{Ce}/^{140}\text{Ce}$  in  $\text{Ce}^{4+}$  vs.  $\text{Ce}^{3+}$  equilibria indicate that this effect nearly cancels the mass-dependent fractionation, leading to a muted overall fractionation. However, the isotope pairs  $^{140}\text{Ce}/^{138}\text{Ce}$  and  $^{140}\text{Ce}/^{136}\text{Ce}$  are expected to show dominantly mass-dependent fractionation, favoring more massive isotopes in  $\text{Ce}^{4+}$ -bearing species.

[1] Lee S-G and Tanaka T (2021) *Geochemical Journal* 55:e9-e17.

[2] Hu JY, et al. (2021) *Science Advances* 7:eabc2962.