

# Acidification of northern New England lakes from rising anthropogenic-sourced atmospheric carbon dioxide concentrations

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We evaluated the impact of rising atmospheric CO<sub>2</sub> (i.e., pCO<sub>2</sub><sup>atm</sup>) due to anthropogenic activities on pH, alkalinity, and acidity of 18 lakes from the north-eastern United States, including Champlain, Winnepesaukee, and Moosehead lakes, using a polythermal, sliding activity/fugacity reaction path model. Atmospheric CO<sub>2</sub> was forced assuming two scenarios from the IPCC's Sixth Assessment Report: 1) scenario SSP2-4.5 in which pCO<sub>2</sub><sup>atm</sup> attains a concentration of 600 ppm in 2100; and 2) scenario SSP5-8.5 in which pCO<sub>2</sub><sup>atm</sup> attains 1100 ppm in 2100 [1]. Equilibrium modelling suggests that most of the studied lakes are supersaturated with respect to CO<sub>2</sub> and are thus local sources of CO<sub>2</sub> to the atmosphere. Nonetheless, as pCO<sub>2</sub><sup>atm</sup> increases, the pCO<sub>2</sub><sup>water</sup> will also increase proportionally to maintain steady-state in the lakes and keep the lake-to-air CO<sub>2</sub> flux constant [2]. Consequently, the lakes will therefore acidify. The modelling suggests that pH will decrease, on average, 0.15 pH units under scenario SSP2-4.5 and 0.32 pH units for scenario SSP5-8.5. These changes correspond to a 31% and 65% increase in the hydrogen ion activity, respectively. Furthermore, the activity of CO<sub>3</sub><sup>2-</sup> is predicted to decrease, on average, by 24% and 49%, respectively for SSP2-4.5 and SSP5-8.5, whereas the saturation state with respect to aragonite would decrease by 21% and 45%, respectively. The predicted changes are like those expected for the Great Lakes [2] and the ocean [3], and consequently may impact lacustrine calcifying organisms. Changes in aluminium (Al) speciation resulting from lake acidification are also evaluated and discussed within the context of Al toxicity to aquatic organisms.

[1] Arias et al. (2021) 6<sup>th</sup> Assessment Report, IPCC, 33-144  
[2] Phillips et al. (2015) *Oceanography* 28 136-145. [3] Orr et al. (2005) *Nature* 437, 681-686.