

Biogenic minerals formation by an Fe^{III}-reducing *Desulfitobacterium* sp. isolate

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Dissimilatory iron reduction is an anaerobic respiratory pathway, wherein ferric (Fe^{III}) reducers couple the oxidation of organic acids, sugars and aromatic hydrocarbons to the reduction of Fe^{III}-species [1]. This may lead to the formation of minerals such as magnetite (Fe^{II}Fe^{III}₂O₄) and siderite (Fe^{II}CO₃) [2], which, in turn, can mediate the reduction of soluble pollutants as pertechnetate (Tc^{VII}O₄⁻) to insoluble oxides (Tc^{IV}O₂) [3].

The genus *Desulfitobacterium* contains obligate anaerobic bacteria that are capable of utilizing a wide range of electron acceptors, including nitrite, sulfite, metals, humic acids and halogenated organic compounds [4].

In this work, the Fe^{III} reduction of a *Desulfitobacterium* species was examined. The microorganism has been isolated from bentonite, which is potentially used as geotechnical barrier in deep geological repositories for radioactive waste [5].

The cultivation conditions included DSMZ 579 medium with Na-acetate as electron donor to reduce Fe^{III} citrate [6]. During cultivation, the formation of white precipitates was observed. The phases were collected both under aerobic and anaerobic conditions and repeatedly investigated by using Raman microscopy and powder X-ray diffraction (pXRD). It was noticed that the phases turned immediately to blue-greenish overnight under oxic conditions. Both Raman spectra and pXRD diffractograms can be attributed to vivianite (Fe^{II}₃(PO₄)₂). Moreover, Raman spectra revealed the possible presence of pyrite (Fe^{II}S₂), siderite, magnetite and hematite (Fe^{III}₂O₃). These results suggest the ability of the bacterium of forming different Fe^{II}-minerals. Notwithstanding, both methods indicate the change of the chemistry of the precipitates according to environmental factors. The Fe^{II}-minerals formation by this microorganism depending on Fe^{III}-compounds and background electrolytes is currently ongoing. The biogenic ferrous minerals will be studied regarding the reduction of Tc^{VII}O₄⁻.

The authors acknowledge the German Federal Ministry of Education and Research (BMBF) for the financial support of

NukSiFutur TecRad young investigator group (02NUK072).

[1] Lovley, 1993, *Annual Review of Microbiology*, 47:263-290

[2] Lee, *et al.*, 2007, *Geomicrobiology Journal*, 24:1, 31-41

[3] Lloyd, *et al.*, 2000, *Appl Environ Microbiol.*, 66(9):3743-9

[4] Villemur *et al.*, 2006, *FEMS Microbiol Rev.* 2006, 30(5):706-33

[5] Drozdowski J., *et al.*, 2018, *HZDR Annual Report*, pp. 40, ISSN 2191-870

[6]

https://www.dsmz.de/microorganisms/medium/pdf/DSMZ_Medit

Accessed on 27.02.23