

Lithium isotopic constraints on widespread clay authigenesis after the Marinoan glaciation

YISHENG YIN¹, GUANGYI WEI¹, PROF. PHILIP POGGE VON STRANDMANN², MAXWELL LECHTE³, DR. SIMON V. HOHL⁴, FEIFEI ZHANG¹, TERRY ISSON⁵ AND HONG-FEILING¹

¹School of Earth Sciences and Engineering, Nanjing University

²London Geochemistry and Isotope Centre (LOGIC), Institute of Earth and Planetary Sciences, University College London and Birkbeck, University of London

³McGill University

⁴Tongji University

⁵University of Waikato (Tauranga)

Presenting Author: ysyin@smail.nju.edu.cn

The Marinoan glaciation stands for the second Snowball Earth glaciation during the late Neoproterozoic. A widespread geological record of cap carbonate marks a transition from an extreme icehouse to greenhouse climate following raised atmospheric CO₂. However, the consequence for such extreme climatic transition remains debated. Here with a sequential extraction approach, we determine the lithium isotope composition ($\delta^7\text{Li}$) of the carbonate and silicate fractions in cap carbonates. We find that carbonate-hosted silicate is dominated by marine authigenic clays with minor contributions from terrestrial materials. Reconstruction of seawater $\delta^7\text{Li}$ from carbonate-hosted silicates suggests a heterogeneous marine Li reservoir in a highly stratified post-glacial ocean, and then rapid mixing of seawater after the Marinoan glaciation. By modelling the $\delta^7\text{Li}$ data, we find that low silicate weathering intensity and high denudation rates (detailed information about weathering regime see [1]) likely characterized the post-glacial environment. In addition, the $\delta^7\text{Li}$ of carbonate minerals resembles that of carbonate-hosted silicate, showing specific effects of silicate component on bulk carbonate $\delta^7\text{Li}$ values, and then emphasizing the significance of component-selective analysis of ancient carbonates for paleo-weathering reconstruction.

Reference

[1] Pogge Von Strandmann et al., 2020, Lithium and Lithium Isotopes in Earth's Surface Cycles. *Elements* 2020; 16 (4): 253–258.