

Molecular Distributions and Stable Carbon Isotopic Composition of Diacids, Oxoacids, and α -dicarbonyls in PM_{2.5} from Underground Coal Fire Field in Inner Mongolia, North China: Implications for Anthropogenic Origin of Azelaic Acid

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To understand the characteristics of chemical components and diacids, oxoacids and α -dicarbonyls in atmospheric aerosols at underground coal fire environment, we collected fine aerosols (PM_{2.5}) from Wuhai, Inner Mongolia, North China, where the underground coal fire occurs severely. PM_{2.5} samples were analyzed for the measurements of carbonaceous components, stable carbon isotopic composition ($\delta^{13}\text{C}$) of total carbon (TC), and molecular distribution and compound-specific $\delta^{13}\text{C}$ of diacids, oxoacids and α -dicarbonyls. Oxalic acid (C₂) was the most abundant diacid species, accounting for 44-63% (avg. 51%) of total diacids, followed by phthalic acid (Ph) (14%). Azelaic acid (C₉) was the third most (7%) abundant diacid, followed by succinic (C₄) and malonic (C₃) acids. The correlations between C₂ and ωC_2 , ωC_3 and C₃ were significant, but not with SO_4^{2-} , which indicate that C₂ was mainly derived by *in-situ* secondary formation and/or primary emission from the underground coal combustion at local scale and less aged. C₉ diacid showed a significant correlation with C₂, C₃, ωC_2 and ωC_3 as well as with Ph acid and C₃/C₄ mass ratios. Such correlations demonstrate that C₉ diacid should have been mainly derived from *in-situ* secondary formation from its precursors emitted from underground coal fires. In addition, $\delta^{13}\text{C}$ of C₉ ($-27.0 \pm 0.9\%$) was comparable with that of Ph acid and similar to the $\delta^{13}\text{C}$ of organic matter in coals, which support its origin of underground coal fire. However, further research is needed to confirm the relative importance of coal combustion for C₉ diacid loading in atmospheric aerosols.