

Natural hydrogen genesis and pyritization

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While most of the known reactions are usually associated with minerals reacting with water under oxidizing conditions (e.g. serpentinization) and radiolysis, the role of pyritization has been underestimated. We aim to explain the H₂ genesis in a sulfur environment and to list proxies helping to localize hydrogen in Earth's crust (formation and accumulation). The used thermodynamic simulations enable to predict the direction of a range of possible metamorphic reactions regarding Gibbs free energies as a function of depth in the crust, temperatures up to 1500°C, pressure up to 10kbar and including the relative partial pressures of H₂ and H₂S gases. We show that iron-based oxide precursors under hydrogen sulfide lead to H₂ formation and pyrite FeS₂ in the upper part of the continental crust (down to 12km using the thermal gradient of 30°C at Qr=1). As the crust can be considered as a semi-open system for gas mobility (lowering H₂ partial pressure by leaking), its genesis should occur at deeper levels. The slow cooling of continental crust through time in cratonic areas is also compatible with the enhancement of exothermic pyritization reactions. The pyritization induces a reduction of ferric into ferrous iron, completing the iron redox cycle to the inverse oxidation process represented by the serpentinization[1]. Pyritization and serpentinization are complementary reactions and can co-exist (e.g. pyritization in acidic black smokers and serpentinization in basic white smokers). Active sites (e.g. black smokers, geothermals) with iron sulfides are therefore propitious areas for H₂ formation. To localize the kitchen of H₂ formation and its accumulation, direct and indirect methods are available such as studying magnetic anomalies, thermal gradient anomalies, gravimetry, fluids and minerals analysis, natural radioactive emissions. To estimate the cost for its exploration, monitoring the H₂ fluxes and localising its possible accumulation via geological analysis of the field is necessary. Some case studies will be discussed in terms of potential investment for using natural H₂ as a clean energy for fuel cells and other industrial

applications.

Reference:

[1] Arrouvel & Prinzhofer (2021) International Journal of Hydrogen Energy 46, 18780-18794

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