

# Computational Chemistry Applied to Solutions: Anions

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Ab initio computational chemistry has been employed in the past by the author to interpret the vibrational spectra of cations such as  $\text{Li}^+$ [1],  $\text{Cd}^{2+}$ [2],  $\text{Mg}^{2+}$ [3],  $\text{Zn}^{2+}$ [4],  $\text{Sc}^{3+}$ [5],  $\text{Al}^{3+}$ [6],  $\text{Ga}^{3+}$ [7],  $\text{In}^{3+}$ [8],  $\text{Bi}^{3+}$ [9], and  $\text{Be}^{2+}$ [10],  $\text{Hg}^{2+}$  and  $\text{Tl}^{3+}$ [11],  $\text{Cu}^+$ [12], and  $\text{Pb}^{2+}$ [13]; anions such as  $\text{SO}_4^{2-}$ [14],  $\text{PO}_4^{3-}$ [15],  $\text{HPO}_4^{2-}$ [16],  $\text{H}_2\text{PO}_4^-$ [17],  $\text{ClO}_4^-$ ,  $\text{BrO}_4^-$ ,  $\text{SeO}_4^{2-}$ ,  $\text{AsO}_4^{3-}$ ,  $\text{VO}_4^{3-}$ [18], the borates[19], and  $\text{HSO}_4^-$ [20]; and ion pairs/complexes such as  $\text{LiX}$  ( $X = \text{F} - \text{I}$ )[21],  $\text{ScCl}_m^{(3-m)+}$  ( $m = 1 - 6$ )[22],  $\text{ZnCl}_m^{(2-m)+}$  ( $m = 1 - 6$ )[23],  $\text{ZnBr}_m^{(2-m)+}$  ( $m = 1 - 6$ )[24], and  $\text{CuCl}_m^{(m-1)-}$  ( $m = 0 - 6$ )[25], in aqueous solution.

In this presentation, the author will discuss the microhydration of selected anions and illustrate the relationships between the vibrational spectra of polyatomic anions in the gas-phase and aqueous-phase.

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