Investigating multiple mine tailings for enhanced weathering methods and CO₂ sequestration

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Anthropogenic CO₂ inputs to the atmosphere are a key cause of global warming, and it is now clear that the Paris Agreement target of limiting the global average temperature increase to 1.5°C cannot be achieved without CO₂ removal from atmosphere on the order of tens of gigatonnes (Gt) per year by 2100. Attention is turning to accelerated CO₂ removal through enhanced weathering and carbonation of silicate mine tailings. The principle is to accelerate natural chemical weathering, whereby minerals dissolve in a reaction with CO₂ and water to form soluble bicarbonates. These products are stored as alkalinity, or precipitate as carbonate minerals. This natural reaction consumes CO₂ and stores it safely for hundreds of thousands of years, but is a slow process, taking several hundred years or more to occur. Enhanced weathering aims to accelerate the weathering and precipitation processes in such a way that significant amounts of CO2 are removed from the atmosphere on human timescales (up to tens of years).

Recent studies¹⁻² have highlighted the high potential for Gtscale removal through enhanced weathering of both annually produced and historic silicate tailings, but critical uncertainties remain regarding which materials are appropriately reactive, how reactions can be sufficiently sped up and how such approaches can be upscaled for industrial implementation. In this study, we investigate the reactivity of a range of silicate mine tailings with CO₂ and water. The sample set, derived from mines distributed globally, includes materials considered to be chemically favourable¹⁻² (e.g., diamond kimberlites, olivine dunites), materials that may be promising¹⁻⁵ (e.g., Al red muds, borates, Cu volcanics) and materials of unknown/minor promise (e.g., lime-treated Cu tailings, SO₂-processing Al products). The aim is to identify the tailings that show suitable potential for CO₂water-rock reactions for reactor schemes with implemented enhanced weathering methods for Gt-scale CO₂ removal.

¹ Bullock et al. (2021) Front. Clim., 3, 694175

² Bullock et al. (2022) Sci. Total Environ., 808, 152111

³ Jones et al. (2006) Environ. Chem., 3(4), 297-303

⁴Guo et al. (2011) Environ. Sci. Technol., 45(11), 4802-4807

⁵ Srivastava et al. (2022) J. Environ. Chem. Eng., 10(2), 107338

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