Major and trace element compositions of partial melts of hydrous pyroxenites and their relationship to volcanism and mantle evolution

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Melting experiments on pyroxenites rich in the hydrous minerals phlogopite (PHL) and K-richterite (KR), some including 5% of the accessory phases ilmenite, rutile or apatite, have been conducted at 15 and 50 kbar to characterise the compositions of initial melts in the continental lithosphere. K-richterite melts rapidly and completely within 50ËšC of the solidus (\approx 1000ËšC @ 15kbar; \approx 1200ËšC @ 50 kbar), resulting in lamproitic melt compositions dominated by the chemical components of KR. Melts of mica pyroxenites without KR are richer in CaO and Al₂O₃. The presence of accessory phases produces melts exceptionally rich in TiO₂and/or P₂O₅, unlike any melts seen at the Earth's surface.

Melting temperatures for CPX+PHL+KR mixtures are close to cratonic geotherms at depths >130km so that minor perturbations of the stable geotherm may rapidly lead to 20% melting. These melts are mobile and effective agents of mantle metasomatism at temperatures well below the melting point of anhydrous peridotite. Trace element analyses of melts indicate that both Tioxides and apatite are essential in the source to reproduce the enriched trace element patterns of lamproites: apatite for the REE and Ti-oxides for Nb, Ta, Zr and Hf. CPX+PHL+KR mixtures have higher bulk partition coefficients for all of the first row transition elements except zinc, with both PHL (9.2) and KR (12) having D_{Ni} comparable to olivine.

Flow-through experiments in which a lamproite melt similar to those produced in the current experiments was reacted with peridotite demonstrated that the melt loses its lamproitic chemistry after reaction with peridotite [1]. The occurrence of lamproites at the surface therefore requires channelled flow that prevents reaction with peridotite.

The presence of orthopyroxene in hydrous pyroxenites assists the production of melts with high K_2O/Na_2O with higher normative leucite contents. These rocks are expected to be common in subduction channels as products of reaction between subducted sediments and peridotite [2]; melts of this composition may be an indication of the involvement of subducted sediments in the source.

[1] Foley, S.F., Pertermann, M. (2021) Geosciences 11, 432.

[2] Sekine, T., Wyllie, P.J. (1983) Journal of Geology 91, 511-528.