

Detecting low levels of radioactive Sr in environmental samples using RPQ- TIMS

SHIGEYUKI WAKAKI¹, JO AOKI², KATZ SUZUKI³,
TAKASHI MIYAZAKI¹, JENNY ROBERTS⁴, HAUKE
VOLLSTAEDT⁴, YOSHITAKA TAKAGAI², DARREN
TOLLSTRUP⁴ AND SATOSHI SASAKI⁴

¹JAMSTEC

²Fukushima University

³Japan Agency for Marine-Earth Science and Technology

⁴Thermo Fisher Scientific

Presenting Author: darren.tollstrup2@thermofisher.com

In some environments, such as sites that have experienced radioactive contamination from nuclear accidents or weapons testing, radioactive strontium (⁹⁰Sr; half-life of 28.8 years) can be found. The monitoring of ⁹⁰Sr is important from a human perspective because when contaminated food or water is ingested, ⁹⁰Sr is concentrated primarily in bones and bone marrow, resulting in bone cancer, cancer of nearby tissues and leukemia. Typically, the abundance of ⁹⁰Sr has been measured using radiometric techniques, such as solid/liquid scintillators or gas ionization detectors. The major disadvantage of this technique is the total analysis time (5-20 days)[1]. In recent years, thermal ionization mass spectrometry has provided a means to dramatically reduce analysis time, improving sample through-put to 10-15 samples/day [2], [3]. However, the peak tailing from the highly abundant ⁸⁸Sr has limited the ability of conventional mass spectrometry to be able to resolve less radioactive samples. Here, we demonstrate the ability of the Retarding Quadrupole Lenses (RPQ) of the Thermo ScientificTM TritonTM Series TIMS to precisely and accurately measure ⁹⁰Sr/⁸⁸Sr ratios as low as 9.1×10^{-12} [4]. This allows the measurement of low activity environmental samples from the vicinity of the Chernobyl and Fukushima nuclear disasters with a modern ⁹⁰Sr activities ranging between 14.8 and 110 Bq/kg.

[1] X. Hou and P. Roos, *Analytica Chimica Acta*, vol. 608, no. 2, pp. 105–139, Feb. 11, 2008. doi: 10.1016/j.aca.2007.12.012.

[2] W. Bu, Y. Ni, G. Steinhäuser, W. Zheng, J. Zheng, and N. Furuta, *Journal of Analytical Atomic Spectrometry*, vol. 33, no. 4, Royal Society of Chemistry, pp. 519–546, Apr. 01, 2018. doi: 10.1039/c7ja00401j.

[3] J. Feuerstein, S. F. Boulyga, P. Galler, G. Stinger, and T. Prohaska, *Journal of Environmental Radioactivity*, vol. 99, no. 11, pp. 1764–1769, Nov. 2008, doi: 10.1016/j.jenvrad.2008.07.002.

[4] S. Wakaki *et al.*, *Scientific Reports 2022 12:1*, vol. 12, no. 1, pp. 1–10, Jan. 2022, doi: 10.1038/s41598-022-05048-7.