Determining controls on hydrogen isotope fractionation in archaeal tetraether lipids in a thermoacidophilic archaeal heterotroph

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Hydrogen from water plays a critical role in nearly all metabolic processes and is incorporated into all biomolecules, often with distinctive isotope fractionations that track processes. The hydrogen isotope composition (δ^2 H) of biomolecules can be used as proxy for past hydrology and ecology. While most research has focused on plant waxes, microbial lipids are of interest because lipid-H can retain its original isotopic composition for up to 10⁸ years [1]. Bacterial lipids, however, are unreliable proxies for paleohydrology because their lipid-water fractionation varies widely (from -150 to +380‰) and their δ^2 H composition mainly reflects central metabolism [2]. The potential for archaeal lipid $\delta^2 H$ to indicate the $\delta^2 H$ of water during growth is understudied. Here, we examine the H isotope fractionation between glycerol dialkyl glycerol tetraether membrane lipids (GDGTs) and water with the model archaeal thermoacidophilic heterotroph, Sulfolobus acidocaldarius, over different pH, temperature, and aeration rates [3]. Separately, we vary the H-isotope composition of the growth water and of the electron donor and carbon source (Dglucose), which allows us to estimate the relative contributions of H from protons in water and organic substrates to the final tetraether lipid products. The ether bonds in harvested GDGTs were chemically cleaved and the resulting biphytanal chains were analyzed for δ^2 H via GC-pyrolysis-IRMS. We show that biphytane δ^2 H is strongly correlated water δ^2 H and all biphytanes are ²H-depleted relative to growth water. We will discuss factors contributing to variation in fractionation factors including culture conditions (i.e., temp, pH, and O₂). Ultimately, these experiments help develop an interpretive framework for the $\delta^2 H$ of archaeal lipid biomarkers in modern and ancient systems that can be validated with environmental samples across similar gradients in major geochemical parameters.

[1] Sessions et al. (2004), *Geochimica et Cosmochimica Acta* 68.7: 1545-1559.

[2] Wijker et al. (2019), *Proceedings of the National Academy of Sciences* 116.25: 12173-12182.

[3] Cobban et al. (2020), *Environmental Microbiology* 22.9: 4046-4056.