

Revealing crystal structure of hexagonal (Fe,Al)OOH_x (HH-phase)

ELENA BYKOVA¹, MAXIM BYKOV², TAKAYUKI ISHII³,
IULIA KOEMETS³, EGOR KOEMETS³, KONSTANTIN
GLAZYRIN⁴, HANNS-PETER LIERMANN⁴, STELLA
CHARITON⁵, VITALI PRAKAPENKA⁵, ALEXANDER
GONCHAROV¹ AND LEONID DUBROVINSKY³

¹Carnegie Institution for Science

²Howard University

³University of Bayreuth

⁴Deutsches Elektronen-Synchrotron

⁵University of Chicago, Argonne National Laboratory

Presenting Author: knilav@gmail.com

Hydrous minerals can deliver water into the deep Earth interiors by subduction and store it from the time Earth was formed. Presence of water can have effect on the Earth's mantle electrical conductivity, melting temperature, viscosity *etc.* Many studies were devoted to the revealing a role of iron oxides (like cubic FeO₂) and their hydrous analogs (such as FeO₂H_x) in the Earth's water cycle and formation of the ultra-low velocity zones. Recently, it was reported that at the conditions of the Earth's lower mantle (107–136 GPa and 2400 K) FeOOH_x can incorporate some amount of aluminum forming a new phase [1]. So-called “HH-phase” was suggested to be a potential candidate for water storage in the deep lower mantle.

Despite potential importance of HH-phase for understanding the Earth's deep water cycle, the information about its crystal structure was absent and chemical composition remain ambiguous. We have synthesized HH-phase by thermal decomposition of (Fe_{0.7}Al_{0.3})OOH starting material at 110 GPa and 2300 K. Based on the single-crystal X-ray diffraction in laser-heating diamond anvil cells we managed to solve and refine the crystal structure of HH-phase, identify its chemical composition and measure bulk compressibility on decompression from 110 GPa. Moreover, a phase with the same crystal structure can be obtained by the thermal decomposition of anhydrous oxides such as Fe₃O₄ and Fe₂O₃. At the same pressure, differences between unit cell volumes of HH-phases obtained from various precursors remain insignificant, suggesting minor content or even absence of water in the studied samples.

[1] Zhang, Yuan, Meng & Mao (2018), *PNAS* 115, 2908-2911.