

Development of an isotope ratio laser spectrometer for rapid and precise clumped isotope measurement: progress and performance

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Time-consuming measurement, large sample size, and assumptions regarding ¹⁷O corrections limit the broad application of carbonate clumped isotope research. We report our recent progress in development of an isotope ratio laser spectrometry instrument based on tunable infrared laser differential absorption spectroscopy (TILDAS). The instrument has three main sections: a gas extraction/dehydration line, a CO₂ sample mixing and delivery system, and a TILDAS dual laser instrument for the isotopologue suite measurement. The gas extraction line follows widespread methods for carbonate-phosphoric acid reaction and then purifies the CO₂ gas by cryogenically removing water vapor and non-condensable gases. The sample introduction system thoroughly mixes CO₂ gas with N₂ gas (which is a unique feature of our laser spectroscopic method) and loads the mixture into the TILDAS instrument for individual measurement of four CO₂ isotopologues (¹⁶O¹²C¹⁶O, ¹⁶O¹³C¹⁶O, ¹⁶O¹²C¹⁸O, and ¹⁶O¹³C¹⁸O). The measurements are compared to a working reference gas. The addition of N₂ gas widens the absorption peaks and improves the high precision measurement of absorption peak areas. Isotopologues associated with ¹⁷O are not present in the peaks measured for the clumped isotope calculation. This automated TILDAS instrument is competitive with the best isotope ratio mass spectrometry (IRMS) systems and surpasses typical IRMS measurements in several key respects, such as rapid and precise measurement (0.01‰ within 45 minutes per carbonate sample, 1 S.E.), small sample size (<20 μmol of CO₂, <2 mg equivalent calcite), and no isobaric interferences. The reported TILDAS Δ_{16O13C18O} values, approximately equivalent to Δ₄₇, show a linear relationship with the theoretical values. Synthetic carbonates grown at controlled temperatures are used to establish an empirical Δ_{16O13C18O}-temperature calibration. Long-term drift is significantly reduced, and one standard deviation of 0.023‰ (n=70) is observed in the measurement of a laboratory carbonate standard (Carrara marble) over a period of 3 months. An automated sample processing system, projected to measure 20 unknowns in a 24 hour period, is nearly completed and we will report on the total throughput and precision of this system.