f-element sorption onto K-feldspar – A comprehensive characterization of mechanism and thermodynamics

JULIA NEUMANN¹, HANNES BRINKMANN¹, SUSAN BRITZ², JOHANNES LUETZENKIRCHEN³, FRANK BOK⁴, MADLEN STOCKMANN⁵, VINZENZ BRENDLER⁶, THORSTEN STUMPF⁷ AND MORITZ SCHMIDT⁸

¹Helmholtz-Zentrum Dresden-Rossendorf ²GRS

³KIT

⁴Institute of Resource Ecology, Helmholtz-Zentrum Dresden-Rossendorf

⁵Helmholtz-Zentrum Dresden-Rossendorf, Research Site Leipzig ⁶Helmholtz-Zentrum Dresden-Rossendorf e.V. (HZDR)

⁷Helmholtz-Zentrum Dresden-Rossendorf, Institute of Resource Ecology

⁸Helmholtz-Zentrum Dresden - Rossendorf e. V. (HZDR)

Presenting Author: j.neumann@hzdr.de

The mobility of radionuclides in the environment, in particular in the context of a deep geological repository for radioactive waste, is heavily influenced by their interactions with charged mineral surfaces. This study investigates the retention potential of feldspars, a main component of granite as one potential host rock for a repository. The focus is on the sorption of trivalent actinides (Am, Cm) and their rare earth analogues (Eu, La, Lu, Nd, Y) as a main source of radiotoxicity in spent nuclear fuel.

A multi-method approach was used, consisting of traditional batch sorption experiments over a broad range of experimental conditions to determine uptake. Generally, retention increases with increasing pH and reaches quantitative retention at near neutral conditions. Furthermore, a spectroscopic study of the sorption structure on the molecular level was conducted. Time-resolved laser-induced fluorescence spectroscopy (TRLFS) using the actinide Cm as a luminescent probe, shows that four surface complexes are formed, an inner sphere sorption complex and its two hydrolysis forms, as well as a ternary feldspar/Cm/silicate complex at alkaline conditions (pH > 10).

Based on the observed comprehensive batch sorption dataset a generic surface complexation model (SCM-A) was developed that describes sorption of trivalent actinides and their rare earth analogues as a function of a variety of geochemical parameters (pH, ionic strength, metal concentration, solid-liquid ratio,...). In a second step, the dataset for the model was further increased by taking the quantitative spectroscopic results into consideration (SCM-B).

The developed SCMs deliver surface complexation parameters of the formed sorption species, which are included in thermodynamic databases. This data is essential for the subsequent calculation of distribution coefficients in modern approaches like the Smart KD-concept[1] and reactive transport modeling. Therefore, this study provides a contribution to a more reliable safety assessment of repositories for radioactive waste. [2]

[1] Stockmann, M. et al., "Smart Kd-values, their uncertainties

and sensitivities - Applying a new approach for realistic distribution coefficients in geochemical modeling of complex systems", *Chemosphere.*, **187**, 277–285 (2017).

[2] Neumann, J. et al., "A comprehensive study of the sorption mechanism and thermodynamics of f-element sorption onto K-feldspar", *J. Colloid Interface Sci.* (2020). https://doi.org/10.1016/j.jcis.2020.11.041.

