## Mica compositions record carbonatite - silicate wall-rock interaction

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Carbonatites contain variable amounts of silicate minerals with mica being the most common one. However, mica abundance in carbonatites should be low because of the low SiO2 and Al2O3 solubility in carbonatitic melts. Enhanced mica formation in carbonatites may therefore record introduction of silica by carbonatite magma - wall-rock interaction. To investigate this further, we studied coarsegrained calciocarbonatite veins of the southern Palabora carbonatite (RSA), as the large carbonatite - silicate wall-rock interface along carbonatite veins preserves such interaction effects particularly well. At the contact between these veins and the host silicate rocks (clinopyroxenite and fenite = metasomatized basement granite/gneiss), a systematic compositional variation in mica correlates with mica abundance reflecting the intensity of contamination. Carbonatitic micas typically prefer to incorporate Mg over <sup>[VI]</sup>Fe<sup>2+</sup>. Accordingly, enhanced mica formation related to contamination by Mg-poor rocks (fenites) causes a prefered Mg consumption from the magma. This results in a local decrease of the Mg/Fe ratio in the magma. The resulting limited availability of Mg finally induces an increased incorporation of [VI]Fe2+ into mica. Since fenite is Fe-poor, its contribution to Fe introduction can be neglected. A direct contamination effect can be observed for Al which is highly abundant in fenite: the stronger the contamination, the higher is the incorporation of Al in mica. On the other hand, contamination by Al-poor host rocks (clinopyroxenites) that supply Si and Mg but insufficient Al promote an enhanced  $^{[IV]}Fe^{3+}$  incorporation into mica and hence the formation of tetraferriphlogopite. This study shows that a small-scale local variation of the mica composition records the intensity and rock type with which carbonatitic magma reacts. We conclude that contamination-controlled compositional changes in mica must be considered when using mica composition as a geochemical monitor of carbonatite evolution.