

## Partitioning of Fe<sup>3+</sup> during peridotite partial melting at 1 bar and 1.5 GPa

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Melt extraction trends in peridotite xenoliths show Fe<sup>3+</sup> behaves incompatibly during partial melting in the mantle [1]. Similarly, Fe<sup>3+</sup> concentrates as an incompatible element in the melt during olivine fractionation in MORB [2], but Fe<sup>3+</sup> does not positively correlate with other incompatible elements, such as Na, in fractionation-corrected MORB, but rather may correlate negatively [2,3]. Cottrell & Kelley [2] hypothesized this trend could result from a decrease in the bulk Fe<sup>3+</sup> partition coefficient ( $D_{Fe^{3+}}$ ) of peridotite with increasing T, but there were insufficient experimental Ds to test this model.

We present the results of multiple-saturation experiments performed on melt + peridotite systems at 1 bar [4] and 1.5 GPa over a range of  $f_{O_2}$  conditions from about QFM-2 to QFM+2 that allow for the measurement of  $D_{Fe^{3+}}$  between spinel, a key host of Fe<sup>3+</sup> in peridotite, and melt. We controlled  $f_{O_2}$  in 1.5 GPa experiments using Pt-Fe alloy capsules of varying composition, which act as sliding redox buffers [5]. We analyzed Fe<sup>3+</sup>/ $\Sigma$ Fe ratios of glasses by XANES and Fe<sup>3+</sup>/ $\Sigma$ Fe ratios of spinels by electron probe using Mössbauer-analyzed correction standards [6].

Spinel-melt  $D_{Fe^{3+}}$  is about a factor of 2 greater in 1 bar experiments performed at 1225 °C with spinel Cr# 0.61-0.68 than in 1.5 GPa experiments performed at 1380-1400 °C with spinel Cr# 0.15-0.33. Although effects of P, T, and X cannot yet be completely deconvolved, the absence of correlation between spl/melt  $D_{Fe^{3+}}$  and spinel Cr# within each set of experiments suggests this effect is related to increasing T and/or P. If so, the spinel contribution to the bulk  $D_{Fe^{3+}}$  can account for a factor of 1.2 change in bulk  $D_{Fe^{3+}}$  between potential temperatures of 1320 and 1450 °C, a significant contribution to the factor of ~2 change need to explain the Na(8)-Fe<sup>3+</sup>(8) trend of MORB [2]. Experimental measurements of pyroxene-melt  $D_{Fe^{3+}}$  are needed to further test this hypothesis. If increasing mantle T leads Fe<sup>3+</sup> to behave more incompatibly, then high Fe<sup>3+</sup> concentrations in primitive MORB will tend to track with high Fe<sup>2+</sup> concentrations, which are already known to increase with potential temperature. Consequently, Fe<sup>3+</sup>/ $\Sigma$ Fe ratios of primitive MORB should vary little as has been observed [2].

[1] Canil et al. 1994, EPSL [2] Cottrell & Kelley 2011, EPSL [3] Bézou & Humler 2005, GCA [4] Davis & Cottrell 2018, Am. Mineral. [5] Kessel et al. 2001, Am. Mineral. [6] Davis et al. 2017, Am. Mineral.