

Melt percolation in the mantle: the message from Mg isotopes

ZHAO-FENG ZHANG¹, XIN-MIAO ZHAO^{2,3}, HUI WANG⁴,
XIN LI¹ AND YA-JUN AN¹

¹ State Key Laboratory of Isotope Geochemistry, Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou 510640, China

² State Key Laboratory of Lithospheric Evolution, Institute of Geology and Geophysics, Chinese Academy of Sciences, P.O. Box 9825, Beijing 100029, China

³ Institutions of Earth Science, Chinese Academy of Sciences, Beijing 100029, China

⁴ University of Chinese Academy of Sciences, Beijing 100049, China

To investigate the effect of melt percolation on magnesium fractionation and mantle Mg isotopic heterogeneity, two suites of well-characterized mantle peridotite xenoliths (magnesium-peridotite and Fe-rich peridotite) and individual minerals as well as associated alkaline basalts from Yangyuan, North China Craton were studied. The magnesium-peridotites ($Mg\# > 89.5$) have relatively homogeneous Mg isotopic compositions ($\delta^{26/24}Mg = -0.26$ to -0.28 , with an average of $-0.27 \pm 0.02\%$; 2SD, $n = 3$), identical to the normal mantle value [1], and they show equilibrium inter-mineral magnesium isotope fractionation between coexisting mantle minerals. By contrast, the Fe-rich peridotites ($Mg\# < 88$), which are strongly metasomatized and enriched in iron, have heterogeneous Mg isotopic compositions, with $\delta^{26/24}Mg$ ranging from -0.38 to 0.03 . Inter-mineral fractionations between co-existing olivine and pyroxene pairs ($\Delta^{26/24}Mg_{\text{opx-ol}}$ and $\Delta^{26/24}Mg_{\text{opx-cpx}}$) are negatively correlated with $Mg/(Mg+Fe)$ of Opx in Fe-rich peridotite, hinting for a possible compositional effect. Our calculations show that kinetic isotopic fractionation caused by diffusion, probably during melt-peridotite reaction is responsible for negatively correlated Mg and Fe or Ca isotopic compositions in these peridotites [2-4].

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[1] Teng (2017) *Reviews in Mineralogy & Geochemistry* **88**, 219-287. [2] Zhao et al. (2017) *GCA* **208**, 368-380. [3] Zhao et al. (2012) *Chem. Geol.* **292**, 127-139. [4] Zhao et al. (2015) *Chem. Geol.* **401**, 96-110.