

## **Ti<sub>10</sub>(Si,P)<sub>6-7</sub> and Ti<sub>11</sub>(Si,P)<sub>10</sub>, new phases from the Luobusa ophiolite, China: Implications for crystallization of Ti-Si-P melts**

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Corundum extracted from chromitite body Cr-11 in the Luobusa ophiolite near Kangjinla (Tibet, China) contains inclusions of super-reduced phases in the Ti-Si-P system, four of which are found in a spheroid 20 μm across, one half of which is composed of TiSi<sub>2</sub> enclosing globules of TiP, the other half comprises two ternary phases for which chemical analysis (EDX) yielded compositions Ti<sub>10</sub>(Si<sub>5.4</sub>P<sub>1.4</sub>) and Ti<sub>11</sub>(Si<sub>7.5</sub>P<sub>2.5</sub>). 3-dimensional electron diffraction collected on the same crystals allowed structure solution and refinement (dynamical) in the hexagonal space group *P6<sub>3</sub>/mcm* with  $a = 7.30(14)$  Å  $c = 5.09(10)$  Å ( $Z = 1$ ) and tetragonal *I4/mmm* with  $a = 9.4(2)$  Å,  $c = 13.5(3)$  Å ( $Z = 4$ ), respectively. The hexagonal phase belongs to the same structural type as mavlyanovite (Mn<sub>5</sub>Si<sub>3</sub>) and the synthetic end-members Ti<sub>5</sub>Si<sub>3</sub> and Ti<sub>5</sub>P<sub>3.15</sub>, but is unique in being a ternary Ti-Si-P solid solution. Cell parameters, symmetry and structure of the tetragonal phase closely resemble binary and ternary synthetic G-phases, e.g., Grytsiv *et al.*, 2006 [1], although no natural or synthetic G-phase has been reported in the Ti-Si-P system. The spheroid could have been a melt droplet that crystallized with decreasing temperature suggested by liquids for the Ti-P [2] and Ti-Si [3] binary systems: Ti<sub>10</sub>(Si,P)<sub>6-7</sub> → Ti<sub>11</sub>(Si,P)<sub>10</sub> → TiSi<sub>2</sub> + TiP. Such melt droplets were products of the interaction of mantle CH<sub>4</sub> ± H<sub>2</sub> fluids with basaltic magmas in the shallow lithosphere (~30–100 km) [4, 5].

- [1] Grytsiv, A. *et al.* (2003) *Intermetallics*, **11**, 351–359. [2] Okamoto, H. (2007) *J. Phase Equilibria Diffusion*, **28**, 587. [3] Fiore, M. *et al.* (2016) *Materials Research*, **19**, 942-953. [4] Griffin *et al.* (2016) *J. Petrology*, **57**, 655–684. [5] Q. Xiong *et al.* (2017) *Eur. J. Mineral*, **29**, 557-570.