

Sulfuric acid weathering in a High Arctic watershed

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Earth's temperature is thought to be regulated by a negative feedback between atmospheric CO₂ levels and chemical weathering of silicate rocks over million-year time scales. Enhanced physical weathering in the Arctic through glacial retreat, permafrost thaw and collapse is exposing and mobilising vast amounts of finely-ground sediment to the agents of weathering. Elevated pyrite mineral oxidation caused by the exposure of sulfide minerals, such as pyrite, during physical erosion can generate sulfuric acid which can weather freshly exposed carbonate minerals and release CO₂ to the atmosphere.

Here we present new coupled oxygen-sulfur isotopic data for a high Arctic River (Zackenberg River, Northeast Greenland) to partition the sources of dissolved sulfate from gypsum and pyrite end members. High frequency water chemistry data from the Zackenberg River shows how these chemical weathering reactions have evolved seasonally and annually over the past 30 years. The extent of release and sequestration of CO₂ from the Zackenberg River watershed is highly dependent on the time of the season, erosional and precipitation events. The compiled datasets demonstrate that the release of CO₂ from the sulfuric acid weathering of carbonate minerals is widespread. The total flux of which can, at times, outweigh that from the long-term sequestration of atmospheric CO₂ via silicate weathering. This scenario could represent a possible modern-day positive climate feedback loop, which is unaccounted for in coupled CO₂-climate models.