Hydrogenation of graphite, diamond, carbonates and iron carbides as the source of hydrocarbons in the upper mantle

I.A. SOKOL^{1,*}, A.G. SOKOL¹, P.A. ZAIKIN², A.A. TOMILENKO¹ AND T.A. BUL'BAK¹

 ¹ V.S. Sobolev Institute of Geology and Mineralogy of RAS, 3 Koptyug Ave., Novosibirsk, 630090 Russia (*correspondence: <u>isokol0x5a@gmail.com</u>)

²N.N. Vorozhtsov Novosibirsk Institute of Organic Chemistry of RAS, 9 Lavrentiev ave., Novosibirsk 630090, Russia

Formation of hydrocarbons by reactions of hydrogenbearing fluids with carbon [1] (13C soot, graphite, or diamond), carbonate-bearing pelites [2] and iron carbides (Fe₃C and Fe₇C₃) [3] was simulated at 5.5-7.8 GPa and 1100-1400°C, fH2 in Pt and Au capsules being controlled at the Mo+MoO₂+H₂O or Fe+FeO+H₂O equilibria. For the first time, formation of hydrocarbons from inorganic compounds was proved by the reaction of ¹³C with hydrogen, which yielded isotopically pure alkanes. The greatest amounts of HCs (CH₄/C₂H₆ < 1, CH₄/C₃H₈ and CH₄/C₄H₁₀ \leq 10) formed at 1400°C in the 10-hr run. The amount of HCs synthesized at 1200°C was twice smaller. The rate of HCs formation was slowest in runs with diamond. At 1200 °C, light alkanes $(C_1 \approx C_2 > C_3 > C_4)$ formed either by direct hydrogenation of Fe₃C or Fe₇C₃, or by hydrogenation of graphite/diamond in the presence of Fe₃C, Fe₇C₃. The CH₄/C₂H₆ ratio in the fluids decreased from 5 to 0.5 with decreasing iron activity and the C fraction increased in the series: $Fe-Fe_3C \rightarrow Fe_3C Fe_7C_3 \rightarrow Fe_7C_3$ -graphite- Fe_3C -magnesite and Fe₃C-H₂O-CO₂ systems at 1200 °C yielded magnesiowüstite and wüstite, respectively, and both produced C-rich Fe₇C₃ and mainly light alkanes (C1=C2>C3>C4). In the experiments containing pelites methaneimine (CH₃N) was found to be the main N-bearing compound. The experiments have provided the first unambiguous evidence that volatile-rich and reduced mantles of terrestrial planets (at fO2 near or below IW) provided favorable conditions for abiotic generation of complex hydrocarbon systems that predominantly contain light alkanes. The conditions favorable for HC formation exist in earth mantle, where slab-derived H2O-, CO2- and carbonate-bearing fluids interact with metal-saturated mantle.

[1] Sokol et al. (2019) *PEPI* **291**, 12. [2] Sokol et al. (2019) *Minerals* **9**, 712. [3] Sokol et al. (2020) *Minerals* **10**, 88. The research was performed by a grant of the Russian Science Foundation (16-17-10041).