Titanium isotope fractionation in synthetic TiO₂-rich lunar melts

LAURA J. A. RZEHAK $^{1,2^{\ast}},$ Sebastian Kommescher 2 and Raúl O.C. Fonseca 1

 ¹ Institut für Geologie, Mineralogie und Geophysik, Universitätsstraße 150, 44780 Bochum (*correspondence: Laura.Rzehak@ruhr-uni-bochum.de)

² Institut f
ür Geologie und Mineralogie, Z
ülpicher Stra
ße 49b, 50674 K
öln

Equilibria between Ti-rich oxides (e.g. ilmenite and rutile) and silicate melt during basalt petrogenesis are known to lead to Ti isotope fractionation in both lunar and terrestrial samples [1,2]. Oxides are typically isotopically light and coexisting silicate melt is isotopically heavy. However, at fO₂ relevant to lunar magmatism (~ IW-1) 10% of the overall Ti is Ti³⁺ [3], which might influence Ti stable isotope fractionation to an unknown extent. In order to determine the effect of Ti-oxides and fO₂ on Ti stable isotope fractionation, experiments with a TiO₂-rich composition were carried out at different fO₂ in gas mixing furnaces. Samples were mixed with a Ti double-spike, and then processed in a highly modified HFSE chemistry to separate Ti from matrix, to determine its Ti isotope composition.

Our results show that from oxidised to reduced conditions (air to IW-1), armalcolite displays light Ti isotope compositions (-0.056‰ to -0.003‰ relative to OL-Ti [4]), whereas silicate melt (+0.035‰ to +0.197‰) and orthopyroxene (+0.033‰ to +0.247‰) show intermediate to high δ^{49} Ti. Our results show that Ti isotope fractionation during mare basalt petrogenesis is likely to be redox dependent and pre-existing mineral-melt fractionation factors may not be directly applied in the case of lunar basalt petrogenesis.

[1] Millet *et al.* (2016) *EPSL* 449, 197-205. [2] Deng *et al.*(2019) *PNAS* 116 (4), 1132-1135. [3] Leitzke *et al.* (2018) *Contrib. to Mineral. Petrol.* 173, 103. [4] Millet and Dauphas
(2014) *JAAS* 29, 1444.