

Calcium complexation by organic matter in calcareous sediments

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Soils and sediments host large stores of organic carbon which can be released to the atmosphere upon mineralization. Ca has the potential to play a key role in preventing mineralization of this organic carbon by promoting mineral protection. It has long been recognized that Ca forms cation bridges that link together negatively charged functional groups from organics and mineral surfaces, stabilizing organic carbon via mineral sorption. However, there is little direct experimental observation of the coordination between Ca, organic matter (OM), and minerals in sediments, which can be used to determine the molecular-scale mechanism by which Ca bridges OM to mineral surfaces.

In the current work, we examined whether and how Ca was associated with OM in calcareous sediments. We sought to determine whether Ca was co-associated with OM and specific minerals, such as iron oxides or clays; as well as what types of OM Ca was associated with. Additionally, we sought to determine the local coordination environment of Ca complexed by OM to assess whether it participated in bridging interactions.

To this end, sediments were subjected to density fractionation to separate organic and mineral components of the sediments. This technique yielded a particulate organic matter fraction, a fraction containing organic-mineral aggregates, and a fraction that was mostly mineral, with a small amount of organics. In each fraction, the co-localization of Ca with different organic species and with Al, Si and Fe minerals was probed using nano secondary ion mass spectrometry (NanoSIMS) and scanning transmission X-ray microscopy (STXM) at the scale of *ca.* 100 nm. Ca speciation was probed further using (micro-)X-ray absorption near edge structure (XANES) and extended X-ray absorption fine structure (EXAFS) spectroscopy. Our analysis sheds new light on the role that Ca plays in stabilizing OM in calcareous sediments, thereby limiting its mineralization and release as CO₂ into the atmosphere.