## What can isotopes tell us about the atmospheric H<sub>2</sub> cycle?

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At a mole fraction of about 500 ppb, molecular hydrogen (H<sub>2</sub>) is the second most abundant reduced gas in the atmosphere after methane. It is important for atmospheric chemistry, and an indirect greenhouse gas due to its reaction with OH, which increases the lifetime of methane. Atmospheric H<sub>2</sub> is produced by incomplete combustion processes, together with CO, and from atmospheric oxidation of CH<sub>4</sub> and non-methane hydrocarbons. H<sub>2</sub> is also emitted into the atmosphere by microbial sources (fermentation, N<sub>2</sub> fixation) and by geologic sources (seeps, volcanoes). The main sinks for atmospheric H<sub>2</sub> are uptake by soil microbes and reaction with OH radicals.

The isotopic composition of H<sub>2</sub> ( $\delta$ D) has been used to investigate its atmospheric budget, based on the fact that main sources, sinks and chemical processes of H<sub>2</sub> have specific isotopic signatures. Partially due to these studies, the atmospheric H<sub>2</sub> cycle is currently relatively well understood.

We will present an overview on the current knowledge on the atmospheric H<sub>2</sub> isotopic budget, and on the remaining questions, information gaps, and potential for future studies. We will also introduce the newly developed clumped isotope measurements ( $\Delta$ DD), and will discuss their potential for studying atmospheric, (micro)biologic and geologic processes.