

Plutonium Immobilization or Mobilization: The Contribution of Microbial Products and Cells

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Plutonium (Pu) redox cycling is known to occur in stratified lakes and ponds, with higher concentrations occurring in the anoxic layers¹. This increase of Pu in the water column likely involves Pu re-mobilization from the sediment and Pu immobilization due to complexation with organic carbon and iron minerals. However, the contribution of microorganisms to the environmental fate and transport of Pu is little understood. In particular, microbial intracellular and extracellular (exudates) products likely provide various metal binding sites for either mobilization or immobilization of actinides. To decipher the role of microorganisms in the biogeochemical cycling of Pu in aquatic systems, a combination of field and laboratory studies were conducted at Pond B, Savannah River Site (SRS, South Carolina, USA).

Pond B is a man-made, monomictic reservoir that received reactor cooling water from R reactor between 1961–1964 containing trace amounts of ^{238,239,240}Pu, ²⁴¹Am, and ¹³⁷Cs. Analysis of monthly water column profiles from our study confirmed seasonal biogeochemical cycling in the pond, including the distribution of redox-sensitive metals and actinides. We also observed shifts in the bacterial community relative abundances with season and depth. Microcosm experiments under various conditions targeted the growth of specific microorganisms (phototrophs, heterotrophs, iron oxidizers, fermenters, and sulfate reducers), which correspond to the likely metabolisms occurring in Pond B. The metal binding capability of the microbial cells, exudates, and intracellular products were tested and compared to ascertain the contribution to Pu mobility. We observed cell density-dependent Pu sorption that differed with enrichment condition, as well as differential Pu binding to certain molecular weight size fractions of microbial products. These results demonstrate the role of microorganisms in complexing and immobilizing Pu in freshwater aquatic systems.

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1. Sholkovitz, E. R. et al. *Nature* **1982**, 300 (5888), 159.