

## **Helium isotope geochemistry of mantle xenoliths and Cenozoic basalts in Jeju Island, South Korea**

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Jeju Island (South Korea) is an intraplate volcano located at the eastern margin of the Eurasian plate, which has begun volcanic activity 2 Ma ago. Although there have been a lot of geochemical studies on Jeju Island, it is still controversial whether the source of magma is upper mantle or mantle plume [1, 2]. Here we first report helium isotope compositions ( $^3\text{He}/^4\text{He}$ ) in mantle xenoliths and basalts from Jeju Island in order to constrain the source of magma. Major element geochemistry of whole-rocks and minerals analyzed by XRF and EPMA, respectively.  $^3\text{He}/^4\text{He}$  ratios of Ol and Cpx separates were measured by a noble gas mass spectrometer. The mantle xenoliths from SE Jeju are spinel lherzolite with  $^3\text{He}/^4\text{He}$  ratios of Ol ranging from 2.9 to 6.5  $R_A$ . These ratios are generally comparable with  $^3\text{He}/^4\text{He}$  ratios of the Korean peninsula (3.5 – 7.9  $R_A$ ) and the European mantle xenoliths (5.2 – 7  $R_A$ ) [3, 4]. The basalts from the west Jeju Island are tholeiitic to alkaline, showing  $^3\text{He}/^4\text{He}$  ratios of Ol and Cpx phenocrysts from 3.5 to 7.3  $R_A$ , similar to the Jeju mantle xenolith values. It is unlikely that  $^3\text{He}/^4\text{He}$  ratios of the basalts indicate crustal contamination by using the relationship of  $\text{SiO}_2$  and  $\text{K}_2\text{O}/\text{P}_2\text{O}_5$ . Thus, the  $^3\text{He}/^4\text{He}$  ratios of the basalts can provide information on the source of magma. According to the seismic tomography model in Jeju Island, decompression melt could occur at the sub-lithospheric mantle depth by edge-driven mantle convection [5]. Moreover, based on trace element modeling [6], it has been suggested that the mantle source of Jeju basalt is mostly garnet lherzolite and eclogite. Therefore, we suggest that the lower part of SCLM played a pivotal role in the production of magma in Jeju Island.

[1] Choi et al. 2006. *Chem. Geol.* 221:40-64. [2] Tatsumi et al. 2005. *J. Petrol.* 46:523-553. [3] Gautheron and Moreira. 2002. *Earth Planet. Sci. Lett.* 199:39-47. [4] Kim et al. 2005. *Geochem. J.* 39:341-356. [5] Song et al. 2018. *J. Geophys. Res. Solid Earth.* 123:6784-6801. [6] Kim et al. 2019. *Lithos.* 326-327:476-490.