

Effect of initial pH on microenvironmental biomineralization of nitrate reducing bacteria

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Microorganisms play an important role in the biogeochemical cycle of iron, and act as the fundamental driving force of elements geochemistry. The nitrate-reducing bacterium BoFeN1 coupling carbon cycling with iron oxidation, controls the transformation and (im)mobilization of heavy metals by oxidation, reduction and chelation [1]. It was found that lower pH microenvironment around phototrophic Fe(II)-oxidizing bacteria prevent cell encrustation by Fe(III) minerals [2]. But the effect of initial pH on microenvironment of nitrate reducing bacteria BoFeN1 encrusting with Fe(III) precipitates is still unknown.

This study aimed to reveal the effect of initial pH on microenvironment of microbial mineral precipitation. It was found that lower initial medium pH slowed the bacteria growth by prolonging the lag phase, decreasing the biomass, and finally influencing the crystallization degree of Fe(III) precipitates. Optical density, XRD, FTIR and TEM results confirmed that the more biomass and higher crystallization degree of mineralized products with the higher initial medium pH. These findings will better our understanding of the microenvironment gradients for cell iron minerals encrustation and the microscale mechanism controlling the microbial biomineralization. Moreover, we've done the sample preparation for FIB/SEM tomography presently, and the next step is to realize 3-D analysis of mineralized products at different initial medium pH.

[1] Kappler, A.; Straub, K. L. *Molecular Geomicrobiology* 2005, 59, 85.

[2] Hegler, F.; Schmidt, C.; Schwarz, H.; Kappler, A. *FEMS Microbiology Ecology* 2010, 74, 592.