

Experimental exploration of TiN as a possible reservoir for planetary nitrogen

KIERSTIN DAVIAU¹, JUNJIE DONG¹, MATTHEW C. BRENNAN¹, TERRY-ANN SUER¹, REBECCA A. FISCHER¹, VITALI B. PRAKAPENKA²

¹Department of Earth and Planetary Sciences, Harvard, Cambridge, MA 02138, USA

²Center for Advanced Radiation Sources, University of Chicago, Chicago, IL 60637, USA

Estimates of the Bulk Silicate Earth (BSE) nitrogen (N) budget do not match predictions from chondritic estimates. Namely, the C/N ratio of the BSE is high (~40 [1]) as compared to the average chondritic ratio (~21 [2]), implying a low BSE N content. Several hypotheses have been suggested to explain this discrepancy, such as the loss of an early N₂-rich atmosphere [1] or the incorporation of differentiated starting materials later in Earth's accretion [2]. If N is instead more compatible in the mantle than other volatiles such as carbon (C), it may be possible that N is sequestered in stable mantle phases. One possible stable N-containing phase is Osbornite (TiN) [3]. TiN is found naturally in enstatite chondrites [4], and has been observed to form experimentally in MORB samples saturated in N at low fO_2 conditions [5]. Here, we explore the stability of TiN at high pressure and high temperature in a diamond anvil cell, to test whether or not TiN could act as a stable N-bearing compound under planetary mantle conditions. Using synchrotron x-ray diffraction, we measured the thermal equation of state of Osbornite up to pressures of ~70 GPa while heating to 2500 K. We find a preliminary equation of state ($K = 286 (\pm 6)$ GPa and $K' = 4.5 (\pm 0.2)$) that agrees well with previous measurements to 45 GPa at room temperature [6]. We find that the cubic rocksalt phase of TiN remains stable at all conditions considered, meriting further study as a possible N reservoir within planetary mantles.

1. Bergin, E.A., et al., Proc. Natl. Acad. Sci., 2015. **112**(29): p. 8965-8970.

2. Grewal, D.S., et al., Science Advances, 2019. **5**(1): p. eaau3669.

3. Martinez, I. and M. Javoy, Mineralogical Magazine, 1998. **62**: p. 955-956.

4. Rubin, A.E. and B.-G. Choi, Earth, Moon, and Planets, 2009. **105**(1): p. 41-53.

5. Libourel, G., B. Marty, and F. Humbert, Geochimica et Cosmochimica Acta, 2003. **67**(21): p. 4123-4135.

6. Chen, H., et al., Journal of Applied Physics, 2010. **107**(11): p. 113503.