## The influence of the vermiculite template on the morphology of manganese dioxide with enhanced immobilization for thallium

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Vermiculite coupled with manganese dioxide (VER-MnO<sub>2</sub>) was fabricated by synthesizing MnO<sub>2</sub> on the vermiculite (VER) template and used for immobilization of Tl(I). The VER exhibited typical layered structure and MnO<sub>2</sub> possessed a flower-like structure with serious reunion phenomenon, while VER-MnO2 illustrated fish-scale shard spread evenly over a large sheet, suggesting that MnO<sub>2</sub> had triumphantly anchored on the VER template and the aggregation of MnO2 was prevented. The VER Template effect contributed higher BET(298.18m<sup>2</sup>·g<sup>-1</sup>) and degree of oxidation of MnO<sub>2</sub>. VER-MnO<sub>2</sub> showed the highest fixation capacity (144.29 mg·g<sup>-1</sup>) than other two materials in the order VER-MnO<sub>2</sub> > MnO<sub>2</sub> > VER, and there was no risk derived from dissolution of Mn. The mechanism of removal Tl(I) was mainly adsorption, ion exchange, and the oxidizability. Fixed-bed column adsorption experiments demonstrated that VER-MnO<sub>2</sub> can be used for purifying drinking water spiked with Tl (20  $\mu g \cdot L^{-1}$ ) and the effective breakthrough volumes were 900 bed volumes until reaching the maximum limits allowed in drinking water  $(0.1 \mu g \cdot L^{-1})$ . The advantages of high renewability, efficiency, nontoxic, low-cost and environmentally friendly materials confirm the great potential application of VER-MnO2. This study would provide a new thought developing for effective, low-cost and environmentally friendly materials for removal of environmental pollution.

Keywords: vermiculite; manganese dioxide; thallium