Isotopic Composition of Natural Chlorate

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Natural ClO_3^- and ClO_4^- commonly co-occur at a molar ratio ≥ 1 . Terrestrial natural ClO₄ has at least a partial atmospheric source based on its elevated ratios of ³⁶Cl/Cl and non-zero Δ^{17} O values. The isotopic composition of ClO₃⁻ could be useful in understanding the origins of per(chlorate) on Earth and Mars. We developed a method to isolate and purify ClO_3 for isotopic analysis and measured the isotopic composition of ClO₃ isolated from NO₃ rich caliche from the Atacama Desert and southern Death Valley (DV) region of the Mojave Desert. Caliche type materials were extracted in water and ClO_3^- was purified using multiple processes including: solvent extraction, selective precipitation, ion exchange, dialysis, microbial reduction of NO₃, and solid phase adsorbents. ClO₃⁻ stable isotopic composition (δ^{37} Cl, δ^{18} O, and Δ^{17} O) was determined by dual-inlet isotope-ratio mass spectrometry. ClO₃ in DV and Atacama samples had distinct ranges of δ^{18} O and Δ^{17} O values, but similar δ^{37} Cl values (Table 1). Compared to Atacama ClO_4 , Atacama ClO_3 had higher δ^{18} O, similar Δ^{17} O, and much higher δ^{37} Cl values. Compared to DV ClO₄, DV ClO₃ had lower δ^{18} O and Δ^{17} O values but higher δ^{37} Cl values. Our results indicate natural ClO_3 is produced in part by processes involving O_3 , as is natural ClO_4 ; however, the formation mechanisms and(or) subsequent isotopic exchanges or fractionations may affect these compounds differently.

Location	Compound	δ ¹⁸ O, ‰	Δ^{17} O, ‰	δ ³⁷ Cl, ‰
Atacama	ClO ₃ -	+7 to +11	+6	-2 to +1
Atacama	ClO ₄ -	-12 to -2	+7 to +11	-20 to -9
DV	ClO ₃ -	-7 to +1	0 to +3	+1
DV	ClO ₄ -	+3 to +26	+8 to +13	-4 to -1

Tabel 1. Ranges of Isotopic Compositions of ClO₃⁻ and ClO₄⁻.