Hygroscopic Behavior and Chemical composition evolution of Aerosols Generated from Mixture Solutions of 3-Methyl-1,2,3-butanetricarboxylic Acid (MBTCA) and NaCl

L. WU¹, D. TALAGA², X. LI¹, H. K. KIM¹, C. U. RO^{1*}, S. BONHOMMEAU², A. DESMEDT², P. M. FLAUD³, E. PERRAUDIN³, E. VILLENAVE³, S. SOBANSKA^{2*}

¹Inha University, Incheon 22212, Korea (*correspondence: curo@inha.ac.kr, ws19910130@yahoo.com, mixinvisnow@hotmail.com. hkkim1959@gmail.com)

²ISM UMR CNRS 5255, Université de Bordeaux, F-33405 Talence, France (david.talaga@u-bordeaux.fr, sebastien.bonhommeau@u-bordeaux.fr, arnaud.desmedt@u-bordeaux.fr, *correspondence: sophie.sobanska@u-bordeaux.fr)

³EPOC UMR CNRS 5805, Université de Bordeaux, F-33615 Pessac, France (pierre-marie.flaud@u-bordeaux.fr, emilie.perraudin@u-bordeaux.fr, eric.villenave@ubordeaux.fr)

Introduction

Secondary organic aerosols (SOAs) are formed and transformed through complex physico-chemical processes in the atmosphere, which lead to their complex chemical compositions. And the interactions between the organic and inorganic components may alter their hygroscopic properties [1]. Recently, 3-methyl-1,2,3-butanetricarboxylic acid (MBTCA), a later generation product of the monoterpenes, has been proposed as one of the most relevant tracer compounds for biogenic SOA formation [2]. MBTCA was thus selected for this study, allowing to improve our knowledge about its properties.

Results and discussion

In the present work, laboratory generated, micrometer sized, pure MBTCA and NaCl-MBTCA mixture aerosol particles of 4 mixing ratios (molar ratios: 1:1, 2:1, 3:1, 1:2), were examined to observe their hygroscopic behavior and obtain chemical micro-structures using in-situ Raman microspectrometry (RMS). Observation and Raman analysis indicated only mono-sodium tricarboxylate (MSMBTL) can be formed regardless of the mixing ratios. The mixtures under mixing ratios of 1:1 and 1:2 showed no ERH and DRH, while the rest experienced single-stage efflorescence due to heterogeneous nucleation onto the excess NaCl.

[1] Li et al. (2017) *Environ. Sci. Technol* **51** (1), 263-270. [2] Kostenidou et al. (2018) *Environ. Sci. Technol* **52**, 1150-1155.