

Calcium isotope fractionation during mantle melting and Ca isotope composition of Earth's upper mantle

YANG WANG (王阳)^{1,*}, YONGSHENG HE^{1,*}, WENNING LU¹, XUNAN MENG¹, FANGZHEN TENG²

¹ State Key Laboratory of Geological Processes and Mineral Resources, China University of Geosciences, Beijing 100083, China * heys@cugb.edu.cn

² Isotope Laboratory, Department of Earth and Space Sciences, University of Washington, Seattle, WA 98195, USA

Significant Ca isotope fractionation has been observed in terrestrial basalts. However, Ca isotope composition of the upper mantle is currently under controversial, with $\delta^{44/42}\text{Ca}$ ranging from 0.42 ± 0.06 to 0.51 ± 0.02 (2SD) [1-3], and the effect of mantle melting on Ca isotope fractionation remains obscure. To better constrain the scale and mechanism of Ca isotope fractionation during mantle melting, here we report high-precision (0.03‰ on $\delta^{44/42}\text{Ca}$) Ca isotope composition for 39 terrestrial samples including 24 peridotites and 15 basalts (MORBs and OIBs). $\delta^{44/42}\text{Ca}$ of 13 non-metasomatized peridotites range from 0.41 ± 0.04 to 0.56 ± 0.03 , positively correlating with MgO and Mg#. These correlations indicate that the upper mantle tend to enrich heavy Ca isotopes with increasing melt extraction. Accordingly, $\delta^{44/42}\text{Ca}$ of the upper mantle is estimated to be 0.43 ± 0.03 based on the Mg# and MgO of it, which is consistent with Kang et al. (2017) (0.43 ± 0.02). Our modelling suggests that $\Delta^{44/42}\text{Ca}$ between peridotite and melt can be 0.05 at 3 GPa, within the estimate in Kang et al. (2017) (0.05–0.12). In addition, $\delta^{44/42}\text{Ca}$ of a glimmerite sample from Northern Tanzania, as an endmember that represents metasomatic melt, is 0.33 ± 0.02 , indicating that the Ca isotope composition of mantle peridotites can also be significantly modified by metasomatic progresses. $\delta^{44/42}\text{Ca}$ of 14 basalts range from 0.37 ± 0.05 to 0.46 ± 0.04 with an average of 0.41 ± 0.04 (2SD), which is consistent with the average of previously reported values [4]. We note that the OIBs in this study seems to have slightly lighter Ca isotope compositions than MORBs by ~ 0.03 . The lighter isotope compositions of OIBs could be resulted from their lower melting degrees or the heterogeneities inherited from the source.

[1] Huang et al., EPSL, 2010, 292(3-4): 337-344. [2] Kang et al., EPSL, 2017, 474: 128-137. [3] Simon and Depaolo, EPSL, 2010, 289(3-4): 457-466. [4] He et al., GGR, 2017, 41(2): 283-302. # This work is supported by the National Nature Science Foundation of China (41673012).