

## The $^{138}\text{La}$ - $^{138}\text{Ce}$ chondritic reference parameters

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The  $^{138}\text{La}$ - $^{138}\text{Ce}$  isotope system (half-life  $1.02 \times 10^{11}$  years) was introduced in the 1980's and the combination of the  $^{138}\text{La}$ - $^{138}\text{Ce}$  and  $^{147}\text{Sm}$ - $^{143}\text{Nd}$  isotopic system is a unique tool to study the behavior of the LREE, especially because of the relative compatibilities ( $\text{La} < \text{Ce}$  and  $\text{Sm} > \text{Nd}$ ) and the redox sensitivity of Ce, which can occur in 3+ and 4+ state, depending on the redox conditions. In this study, 18 different meteorites (10 carbonaceous, 5 ordinary and 3 enstatite chondrites of petrological types 1-6 and one mesosiderite) were used to better define the Chondritic Uniform Reservoir (CHUR) reference value, allowing direct comparison of terrestrial data to that expected in a chondritic Earth. Therefore, Sm-Nd and La-Ce isotope compositions and concentrations in all meteorites were determined from one aliquot. Elemental concentrations were determined by isotope dilution using a mixed  $^{138}\text{La}$ - $^{142}\text{Ce}$  tracer and isotope compositions were obtained by MC-ICP-MS [1]. The Sm-Nd results are in a good agreement with the literature [2]. Notably, there is a resolvable isotopic difference between different AMES metal batches. For comparison, all data (literature and this study) were recalculated to the Mainz-AMES value of  $1.33738 \pm 1$  (2rsd) proposed by [3]. The mean values of  $^{138}\text{Ce}/^{136}\text{Ce}$  obtained here are not fully consistent with all previous studies [4, 5]. After carefully revisiting all results, we now recommend a new  $^{138}\text{Ce}/^{136}\text{Ce}$  weighted mean for chondrites ( $n=17$ ) of  $1.33685 \pm 3$  (2 se). Importantly, our  $^{138}\text{La}/^{136}\text{Ce}$  values determined by ID-MC-ICP-MS are ca. 10% higher than proposed in a recent study [5] which did not employ isotope dilution for La-Ce. A literature comparison of La/Ce values for rock reference materials confirms a greater accuracy of the isotope dilution approach.

[1] Schnabel et al. (2017) *JAAS*, 32, 2360 [2] Bouvier et al. (2008) *EPSL*, 273, 48-57 [3] Willbold (2007) *JAAS*, 22, 1364-1372 [4] Bellot et al. (2015) *Geochim. Cosmochim. Acta*, 168, 261-279 [5] Willig&Stracke (2019) *EPSL*, 509, 55-65