

Degradation of Cu-bearing uranyl As-P micromineral phases from Krunkelbach uranium deposit, Southern Germany

I. PIDCHENKO^{1,2*}, S. BAUTERNS^{1,2}, L. AMIDANI^{1,2}, K.O.
KVASHNINA^{1,2}

¹The Rossendorf Beamline at ESRF, CS40220, 38043
Grenoble Cedex 9, France. stephen.bauters@esrf.fr,
lucia.amidani@esrf.fr, kristina.kvashnina@esrf.fr

²Helmholtz Zentrum Dresden-Rossendorf (HZDR), Institute
of Resource Ecology, 01328 Dresden, Germany.

(*correspondence: i.pidchenko@hzdr.de)

The abandoned Krunkelbach uranium (U) mine, Southern Germany, with 2-3 km surrounding area represents a unique natural analogue site with accumulation of U minerals suitable for investigations of potential mobilization-immobilization processes expected in a real spent nuclear fuel repository. A specific feature of the site is the occurrence of more than forty secondary U minerals, from mixed redox U oxy-hydroxides to alkaline metal uranyl silicates, thus representing a wide scale of U ore weathering events. In this work the combination of synchrotron and laboratory techniques is used to unveil U speciation and micro heterogeneities in U phases accumulated on granitic rock outcrop from U deposit area. Available data on the age of the secondary U mineralization indicates that oxidizing processes at the site started some 340,000 years ago and continues up to date. Several phases close to $\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_{2-x}(\text{AsO}_4)_x \cdot 8\text{H}_2\text{O}$ are identified on $1 \times 2 \text{ mm}^2$ area with presumably older, more evenly distributed $\text{Cu}(\text{UO}_2)_2(\text{SiO}_3\text{OH})_2 \cdot 6\text{H}_2\text{O}$ and (Fe, Ba, Pb)(UO_2)₂(WO_4) (OH)₄ · 12H₂O, mineralization. Based on a multi-technique investigation 10-200 μm $\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_{2-x}(\text{AsO}_4)_x \cdot 8\text{H}_2\text{O}$ particles with widely varying As-P content are analyzed. The evidences of a degradation occurred on some zones on the selected crystals are found associated with higher As and decreased P content. This observation can be apparently attributed to different degradation properties of the mixed As-P phases depending on As-P ratio and originate from different solubility properties of $\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$ and $\text{Cu}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$ species. The conditions for preferential formation of As rich $\text{Cu}(\text{UO}_2)_2(\text{XO}_4)_2 \cdot 8\text{H}_2\text{O}$ [X=As, P] phases and its possible role on U transport in environment under oxidizing conditions are discussed.