

Dissolution and mobility of neptunium under vadose zone conditions

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Multi-scale Approach to Neptunium Geochemistry

With a long half-life and potentially high environmental mobility, neptunium (Np) geochemistry is a risk-driver in legacy nuclear waste management, but field-scale evidence of Np transport is limited. Field lysimeters have become an important means for evaluating vadose zone transport by simulating natural variability while still maintaining experimental control through well-characterized source materials and careful tracking of solid and aqueous phases over time. Lysimeters are used to probe the effect of oxidation state on neptunium transport from $\text{Np}^{\text{IV}}\text{O}_2$ and $\text{Np}^{\text{IV}}\text{O}_2(\text{NO})_2(\text{H}_2\text{O})_2$ sources, the effect of geochemical conditions and source characteristics on dissolution mechanisms, and the effect of colloids at field scales. $\text{NpO}_2(\text{s})$ data is also compared to $\text{PuO}_2(\text{s})$ lysimeter data, as $\text{NpO}_2(\text{s})$ is an oxidation state analogue for $\text{PuO}_2(\text{s})$.

Analogous laboratory-scale dissolution studies under vadose zone conditions are used to further explain dissolution mechanisms of $\text{NpO}_2(\text{s})$. Solid phase characterization using electron microscopy is emphasized to detect surface alteration. Lab work seeks to confirm and enhance field data for $\text{NpO}_2(\text{s})$ and bridge understanding of $\text{NpO}_2(\text{s})$ dissolution from field to nano scale.

Dissolution Mechanisms of $\text{NpO}_2(\text{s})$

Field data show migration from $\text{NpO}_2(\text{s})$ can be attributed to transport of soluble neptunium ion and colloids of $\text{NpO}_2(\text{s})$. Extent of two-dimensional transport from $\text{NpO}_2(\text{s})$ varies based on depth in the soil column due to changes in volumetric water content. Observed transport from $\text{NpO}_2(\text{s})$ is far greater than from $\text{PuO}_2(\text{s})$ in field lysimeters, but both systems show evidence of colloid transport and both solids remain in the initial oxide form after retrieval. Electron microscopy of $\text{NpO}_2(\text{s})$ reveals significant alteration of the material along grain boundaries. Together, field and lab scale data indicate that dissolution of $\text{NpO}_2(\text{s})$ occurs via alteration of phases along grain boundaries and subsequent transport of colloidal Np.