Mg-Al cation exchange in MgAl₂O₄ spinel and its short-range ordering

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MgAl₂O₄-spinel shows substantial amounts of intersite Mg-Al cation exchange (denoted by the inversion parameter *x*), which modify the structural features and influence the physical and chemical properties. Here we first critically reviewed the kinetics studies on the Mg-Al cation exchange reaction, and then reviewed all thermodynamics experiments. With further necessary temperature correction and data removal, we have obtained a generally reliable *x*-*T* dataset (71 data pairs in total) at ambient *P*. Fitting these *x*-*T* data to three most commonly used thermodynamic models, we have obtained more accurate model parameters. The model-fitting results suggest that the $T \Delta S_D$ might be important. We consequently evaluated the experimental results from different research fields. The same conclusion that the $T\Delta S_D$ should not be neglected has been obtained.

To understand the origin of the $T \Delta S_D$ and constrain its value, we additionally performed theoretical simulations using the DFT technique. Calculations were conducted using the VASP code, with the exchange-correlation interaction approximated by the LDA method. We considered all the possible cation configurations at the *x* values of 0, 0.125, 0.25, 0.375 and 1 (~1000 different configurations). The results clearly show that there is indeed short-range Mg-Al cation ordering, which significantly reduces the residual entropy. Our observation is in good agreement with some reversed phase-equilibrium data in the literature.

Cation order-disorder phenomenon is very common in important rock-forming minerals such as olivine, pyroxene, garnet and ringwoodite. The evidence to the existence of short-range cation ordering observed in the MgAl₂O₄ spinel has profound effects on the thermodynamic properties of these minerals.