

Linking Impurity Incorporation to their Effect on Mineral Growth

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Many minerals show useful capabilities to immobilize toxic heavy metals, e.g. in nuclear waste management or oil/gas extraction. The link between heavy metal incorporation and their effects on growth rates and morphology has not been fully established. Here, we chose the model system of barite (BaSO_4) and Sr to elucidate the relationship between growth and incorporation rate. We combined in-situ hydrothermal growth observations by atomic force microscopy (AFM) with high-resolution ex-situ chemical imaging. To profile the Sr-distribution within the grown solid, we employed atom probe tomography (APT), time of flight secondary ion mass spectrometry equipped with AFM (AFM-ToF-SIMS) and transmission electron microscopy (TEM). Barite growth experiments under incrementally increasing Sr-concentration showed a decrease in growth rate for aqueous $[\text{Sr}]/[\text{Ba}] < 1$, shifting to an increase at aqueous $[\text{Sr}]/[\text{Ba}] > 1$ by a transition from $\langle 120 \rangle$ step growth to $[010]$ step growth. By chemical imaging, we observed the preferential incorporation at high aqueous $[\text{Sr}]/[\text{Ba}]$. For aqueous $[\text{Sr}]/[\text{Ba}] < 5$, we found a correlation between aqueous $[\text{Sr}]/[\text{Ba}]$ and the newly grown solid composition, whereas this was not the case for higher aqueous $[\text{Sr}]/[\text{Ba}]$. We will discuss the results based on current thermodynamic models and demonstrate the need for kinetic process models to interpret our observations.