

Spectroscopic monitoring of the aging of sunscreens with TiO₂ UV filters

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There is an increasing variety of nano-enabled consumer products on the market. Besides food, personal care products are a particularly sensitive category since they come in direct contact with the skin and mucosae, and might even be (accidentally) ingested (e.g. toothpaste). In an effort to make these products benign for both humans and the environment, the strategy aims to address safety issues from the very beginning following the "safe(r) by design" approach. Here we show the implementation of this approach applied to the design of sunscreens. These products contain nano-sized TiO₂ as mineral UV filters. Since, besides being an efficient UV filter, TiO₂ is also a photocatalyst, it is necessary to protect the skin against these harmful effects. This is achieved by applying one or multiple shells around the TiO₂ core with the purpose of blocking photocatalysis and facilitating dispersion of the nanocomposite in the lotion. These nanomaterials need to be safe for human use and benign when released into the environment. Several formulations of the TiO₂ based sunscreens were artificially aged so as to determine and quantify the alteration. To counteract photocatalytic effects, SiO₂ and AlOOH shells were selected to coat the TiO₂ core. These shells behaved very differently: whereas dissolution/degradation of the Al based shell was slow and limited, the SiO₂ shell was far more sensitive to aging and most of the coating was eliminated from the nanocomposite. Spectroscopic data suggest dissolution as the main mechanism.