Silicate vs carbonate weathering: isotopic co-variability in the Fraser River, Canada

K.K. BITTERWOLF^{1*},

B. PEUCKER-EHRENBRINK², A. EISENHAUER³, D.P. SANTIAGO RAMOS⁴, J.A. HIGGINS⁴, AND A. PAYTAN¹

 ¹University of California, Santa Cruz, Santa Cruz, CA 95064, USA (*correspondence: kimberleybitterwolf@gmail.com)
²Woods Hole Oceanographic Institution, Woods Hole, MA, USA
³GEOMAR Helmholtz Centre For Ocean Research Kiel, Germany
⁴Princeton University, Princeton, NJ, USA

Constraining the chemical composition of paleo-oceans inherently informs paleoclimate models by elucidating changes in the atmospheric composition and geologic make-up of Earth's surface. Major ion and isotopic compositions of seawater have varied widely over geologic time, and often in synchronous fashion [1]. However, the underlying mechanisms are not well understood. To improve our understanding, we aim to quantify runoff contributions from silicate and carbonate weathering - two of the major input fluxes of solutes to the global ocean. This goal is being accomplished by examining isotopic co-variation in the dissolved load of the Fraser River (British Columbia, Canada) during periods of more silicate- vs. more carbonate-dominated runoff. The Fraser serves as a natural laboratory for the investigation of these two weathering contributions due to its seasonally-distinct sourcing of runoff, which shifts from predominantly young igneous rocks during base flow to radiogenic (Paleozoic-Precambrian) sediments during the freshet [2]. Five isotope systems relevant to weathering contributions were selected (δ^7 Li, δ^{26} Mg, $\delta^{44/40C}$ Ca, 87 Sr/ 86 Sr, $\delta^{88/86}$ Sr) and analyzed in samples from 2010 - 2012, which were collected as a part of an ongoing time series study of the river's main stem. The results will be used to construct an index of isotopic co-variation during periods of stronger contributions from young igneous bedrock vs. radiogenic sediments, including carbonates.

- 1. H. Vollstaedt *et al.*, *Geochim. Cosmochim. Acta.* **128**, 249–265 (2014).
- B. M. Voss et al., Geochim. Cosmochim. Acta. 124, 283– 308 (2014).